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DMIC Memorandum 201

COMPATIBILITY OF MATERIALS WITH ROCKET  
PROPELLANTS AND OXIDIZERS

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**AD-613 553**

**COMPATIBILITY OF MATERIALS WITH  
ROCKET PROPELLANTS AND OXIDIZERS**

**W. K. Boyd, et al**

**Battelle Memorial Institute  
Columbus, Ohio**

**January 1963**

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DMIC MEMORANDUM 201

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ROCKET PROPELLANTS AND GRIDIZERS

## ERRATA

In Table 6 on page 12, several entries are in error. The corrections are listed below. In table 29, a rating for Monel is supplied that was inadvertently omitted during copying from the worksheets.

## ERRATA FOR TABLE 6

Material	Temperature, F								References
	Gas				Liquid				
	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4	
309 Stainless Steel	500			570					7,207
309 Cb Stainless Steel	500			570					7,207
310 Stainless Steel	500			660					7,207
347 Stainless Steel	390			500	-310		-320		7,120,207,274
430 Stainless Steel		400	390	600					7,207,211
Azmco Iron	390	500	167						7,159,160,207
Iron (0.004 Si)		200	400	390					82,159,160
Iron (0.79 Si)		100	300	390					82,159,160
Sheet Steel	390	660		500					7,207
SAE 1010	100	200	400	>400					82,160
SAE 1011				570					7,207
SAE 1020				390					7,207
SAE 1030	660		390	500					7,207
Music Wire				570					7,207
A-Nickel		1000	750	>1200					7,73,82,120,143,153,159,160,211,274
Monel		1000	750	>1200					7,73,82,120,143,153,159,160,211,274
Inconel		1000		<750					7,207,211
Deoxidized copper				<400					7,207
Brass 70-30		200		400					82,159,160
Magnesium MA (1.2% Mn)	140	200							82,159,160
Magnesium PS-1A	140	200							82,159,160

## ERRATUM FOR TABLE 29

Monel	75	125	100	75	211
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# COMPATIBILITY OF MATERIALS WITH ROCKET PROPELLANTS AND OXIDIZERS

W. K. Boyd, W. E. Berry, and E. L. White\*

## GENERAL CONSIDERATIONS

An important consideration in rocket technology is the compatibility of each propellant with the container material. Serious problems arise because many propellants are extremely reactive and their containment is possible only with a few materials of construction. The resistance of many alloys to fuels and oxidizers is dependent entirely on the formation of an inert, corrosion-resistant film or barrier coating. In addition to corrosion problems, the presence of some metals tends to promote decomposition of the propellant. Also, certain metal/oxidizer combinations may ignite if subjected to impact.

Four years ago DMIC Memorandum 65 was issued and included all information available at that time on compatibility of materials of construction with rocket propellants and oxidizers. Since that time additional information on compatibility has been generated for new, as well as the more established, fuels and oxidizers. This report contains these new data combined with the information presented in DMIC Memorandum 65. In order to expand the usefulness of the report, the source of the data is referenced for each material.

This memorandum summarizes the available information on the compatibility of liquid rocket propellants with prominent materials of construction. It is pointed out that compatibility data for materials not ordinarily covered by the Defense Metals Information Center are included. These data were found during the search for information on materials that are within the scope of the DMIC, and are included for convenience. Fuels and oxidizers of current interest are discussed. The corrosion data which are presented will apply to storing, handling, and control equipment outside of missiles and to missile components excluding combustion chamber. The compatibility of materials with reaction products in combustion chambers, nozzles, etc., has not been considered. Included in the summary are data for many nonmetallic materials. These data were collected in conjunction with those obtained for metals but no concerted effort was made to secure compatibility data for nonmetals.

The memorandum is subdivided into sections according to the propellant. Each material of construction is rated for a given medium as belonging to one of four classes, based primarily upon corrosion resistance. Consideration also is given to such factors as catalytic decomposition and sensitivity to impact.

## CLASSIFICATION OF MATERIALS OF CONSTRUCTION

### Metals

#### Class 1

The Class 1 materials are those which exhibit a corrosion rate of less than 1 mil per year. The material does not promote decomposition of the propellant or oxidizer and is free from impact sensitivity.

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#### Class 2

The materials falling in Class 2 are similar to those in Class 1, except that the corrosion rate may be as great as 5 mils per year.

#### Class 3

A material in Class 3 shows only fair corrosion resistance; rates of attack may be of the order of 5 to 50 mils per year. The material may also cause a moderate breakdown of the propellant, but it is not shock sensitive under conditions likely to be encountered in service.

#### Class 4

Materials in this class are not considered usable for containing the propellant; they have corrosion rates greater than 50 mils per year, cause extensive decomposition of the propellant, cause spontaneous ignition, or react on impact.

These classifications are summarized in Table 1.

TABLE 1. COMPATIBILITY CLASSIFICATIONS FOR METALS<sup>(a)</sup>

Class	Rating	Corrosion Resistance	Decomposition of Propellant	Shock Sensitivity
		Penetration Rate, mils/year		
1	Excellent	<1	No	No
2	Good	<5	No	No
3	Fair	5 to 50	Some	No
4	Poor	>50	Extensive	Yes

(a) The classification of a material is based on the lowest rating of any one of the three properties.

### Nonmetals

Ratings for nonmetals are also somewhat arbitrary but wherever possible they follow those described in the Titan II Storable Propellant Handbook, (169) The classifications are summarized in Table 2.

TABLE 2. COMPATIBILITY CLASSIFICATIONS FOR NONMETALS

	Class			
	1	2	3	4
Volume Change, percent	0 to +25	-10 to +25	-10 to +25	<-10 or >+25
Durometer Reading Change	±3	±10	±10	<-10 or >+10
Effect on Propellant	None	Slight change	Moderate change	Severe
Visual Examination	No change	Slight change	Moderate change	Severely blistered, or cracked, dissolved
General Usage	Satisfactory, general use	Satisfactory for repeated short term use	Satisfactory for short time use	Unsatisfactory

The compatibility data have been tabulated according to the maximum temperature permissible for a given material in Classes 1, 2, and 3. The minimum temperature at which a given material becomes Class 4 (noncompatible) also is listed. For example, a notation of RT under Class 1 means that the material would fall into this classification up to room temperature. It will be noted, in many instances, that no temperature is listed for one of the more resistant classifications. This does not necessarily mean that the material does not fall in Class 1 or 2 at some temperature, but rather that insufficient data are available to assign a temperature limit.

Occasional entries indicate that a material has a higher rating at higher temperatures, e.g., Class 1 at 160 F and Class 2 at 80 F. These entries reflect a conflict in reported data. In many of these cases, it is recommended that the original references be consulted where possible, to determine which results were obtained under conditions most nearly simulating the application in question.

In many cases, a material does not fit into a classification because (1) there is a scarcity of numerical data, or (2) the decomposition effects on the propellant are of primary concern. Hydrogen peroxide is a good example of a propellant for which it is difficult to classify construction materials. In such a case, the classification used in the table is described in detail for the material in question. Many materials are listed by trade names. Similar materials marketed under other trade names probably would be given the same classification. However, only materials for which actual data are available are listed.

#### SOURCES OF INFORMATION

The information on which this memorandum is based came from a variety of sources. A list of references is given at the end of the text. Appropriate references for a particular propellant material are listed at the beginning of each section. References are also included for each individual entry. In addition to reference material obtained from published literature, specialists in companies active in the development of rocket propellants were contacted either by letter, by telephone, or in person. The cooperating companies are listed below.

Aerojet-General Corporation  
Food Machinery and Chemical Corporation  
Becco Chemical Division  
Belmont Smelting and Refining Works, Inc.  
General Dynamics Corporation  
Convair Astronautics Division  
Callery Chemical Company  
Research and Development Division  
Celanese Corporation of America  
Columbia-Southern Chemical Corporation  
Commercial Solvents Corporation  
Douglas Aircraft Company, Inc.  
Diamond Alkali Company  
The Dow Chemical Company  
Texas Division  
E. I. du Pont de Nemours and Company, Inc.  
Foote Mineral Company  
Allied Chemical Corporation  
General Chemical Division  
Hughes Tool Company  
HEF, Inc. - Hooker Chemical Corporation and  
Foote Mineral Company

Hercules Powder Company  
California Institute of Technology  
Jet Propulsion Laboratory  
Lockheed Aircraft Corporation  
Missile Systems Division  
Lithium Corporation of America  
Menasco Manufacturing Company  
Metal Hydrides, Inc.  
Monsanto Chemical Company  
Research and Engineering Division  
Chemetron Corporation  
National Cylinder Gas Division  
Union Carbide Corporation  
Kinde Company  
Union Carbide Metals Company  
National Carbon Company  
Olin-Mathieson Chemical Corporation  
Pennsalt Chemicals Corporation  
Thiokol Chemical Corporation  
Reaction Motors Division  
North American Aviation, Inc.  
Rocketdyne Division  
Rohm & Haas Company  
Stanford Research Institute  
Solar Aircraft Company  
Sinclair Research Laboratories, Inc.  
Titanium Metals Corporation of America  
Virginia-Carolina Chemical Corporation  
Wyandotte Chemicals Corporation.

General information on compatibility and properties of large groups of propellants are contained in References 3, 23, 81, 102, 110, 151, 196, 198, 211, and 237. Information on handling, safety, and toxicity is included in References 81, 195, 291, and 302.

#### AMMONIA (NH<sub>3</sub>)\*

Ammonia is a pungent colorless gas that is alkaline in nature. It can be liquified at room temperature at pressures above 100 psia. The vapor irritates the eyes and respiratory tract. The threshold-limit value of toxicity in the atmosphere is 50 ppm.

Stainless steel, carbon steel, nickel alloys, silver, platinum, gold, and tantalum are sufficiently resistant to anhydrous ammonia to be placed in Class 1, as shown in Table 3. Inconel, gold, platinum, and tantalum are Class 1 materials in moist ammonia. Carbon steel and cast iron are also quite resistant and are normal materials of construction for ammonia service.

The copper alloys are less resistant than steel and have the disadvantage of being susceptible to cracking in ammonia atmosphere.

The upper temperature limit of many metals is related to the initiation of the nitriding process. Inconel is more resistant to nitriding than other nickel alloys, mild steel, or stainless steel.

Many organic materials are suitable for ammonia service. Plastics and elastomers usually resist attack up to their softening point.

Most inorganic construction materials are not attacked by ammonia. Graphitic materials are considered best for handling ammonia gas at very high temperatures.

\*Ammonia: see References 80, 81, 82, 94, 102, 109, 110, 127, 128, 151, 181, 199, 211, 214, 217, 221, 287, and 295.

Hi-Cal-3\*

The composition and properties of Hi-Cal-3 are classified.

Many of the common construction materials are compatible with Hi-Cal-3. Mild steels, stainless steels, copper alloys, nickel alloys, aluminum, titanium, tantalum, and lead can all be used up to 120 F and are listed as Class 1 in Table 4. Above this temperature, no data are available. Organic materials which are compatible with Hi-Cal-3 are listed in Table 4 as Class 2.

Pentaborane ( $B_5H_9$ )\*\*

Pure pentaborane is a clear colorless liquid that possesses an odor similar to that of rotten pumpkin. It is pyrophoric and highly toxic. Maximum allowable exposure is less than 1.0 ppm. It has a vapor pressure of 77 mm at 77 F and boils at 137 F.

Most of the metals used in rocket construction are compatible with pentaborane, including iron, steel, stainless steel, aluminum, copper, brass, magnesium, titanium, etc. Data are presented in Table 5. Teflon, Viton, Kel-F, and fluorosilicon rubber are included among the plastics and elastomers that are compatible with pentaborane. Pentaborane forms shock-sensitive mixtures with most of the chlorinated hydrocarbons that are used as degreasers or solvents.

Trialkyl Boranes\*\*\*

Materials which have withstood 1 month's exposure in triethylborane [ $(C_2H_5)_3B$ ] (a colorless liquid that boils at 203 F) at 160 F with no apparent attack include:

Metals	Nonmetals
Mild steel	Teflon
Stainless steel	Phenolite
Aluminum	Garlock 900 packing
Brass	Garlock red rubber
Nickel	Koppers 6200
Monel	Super Dylan polyethylene
Inconel	
Lead	
Copper (pitted)	

Materials which have withstood 2 weeks' exposure in tri-n-butyl borane [ $(C_4H_9)_3B$ ] (a colorless liquid that exerts a vapor pressure of 20 mm at 228 F) at 122 F with no apparent attack include:

Metals	Nonmetals
Mild steel	Teflon gasket
Stainless steel	Palmer to gasket
Aluminum	Koroseal
Brass	Kel-F
Nickel	Hycar
Copper	Nylon gasket

\*Hi-Cal-3: see Reference 103.

\*\*Pentaborane: see References 56 and 294.

\*\*\*Trialkyl boranes: See Reference 57.

A number of fluorine compounds are being considered as oxidizers for rocket-propulsion systems. These include fluorine, chlorine trifluoride, bromine trifluoride, bromine pentafluoride, iodine pentafluoride, oxygen difluoride, oxygen difluoride-oxygen mixtures, perchloryl fluoride, perchloryl fluoride-tetrafluoro-hydrazine mixtures, and fluorine-oxygen mixtures (FLOX). All of these materials are extremely active chemically. Under the proper conditions, they will react with almost every known element; hence, they present severe corrosion and compatibility problems. These oxidizers, however, are not for the most part susceptible to thermal and catalytic breakdown and present little or no problem in this respect. Therefore, compatibility ratings are based primarily on the reaction of the medium with the material in question.

Fluorine ( $F_2$ )\*

Fluorine is a yellowish gas that has a pungent odor and is irritating to the respiratory tract. The threshold limit of toxicity of fluorine in the atmosphere is 0.1 ppm. It is normally handled as a liquid at -310 to -320 F.

Many metals perform well in liquid and gaseous fluorine as can be seen from the data in Table 6. It is believed that a protective fluoride film which forms on the surface of most metals imparts corrosion resistance. On the other hand, some experiments have revealed no increased corrosion on specimens immersed in liquid fluorine and wire brushed to remove any film, suggesting no passivating effect from metal fluorides. (275)

Traces of water in the system react with fluorine to form hydrofluoric acid. This acid tends to attack some of the materials which are resistant to uncontaminated fluorine. Since moisture may be present in many systems, Monel is usually chosen as a construction material because, in addition to being resistant to fluorine, it also is resistant to hydrofluoric acid.

Titanium ignites\*\* in liquid fluorine if subjected to impact or rupture. However, ignition does not propagate as it does in liquid oxygen. Although ignition has occurred in at least one specimen of aluminum alloys, aluminum is not considered to be impact sensitive in liquid fluorine. (275) Dynamite cap explosions against tubes filled with liquid fluorine have failed to cause ignition in Monel, nickel, copper, brass, 304 stainless steel, 316 stainless steel, 347 stainless steel, and 1100 aluminum. (275) Tensile tests to fracture in liquid fluorine have not caused ignition of AM350, 304L stainless steel, 301 stainless steel, ASM 6434, 2014-T6 and 7075-T6 aluminum, Inconel X, and Ti-6Al-4V. (233)

\*Fluorine: see References 7, 14, 15, 16, 20, 40, 52, 58, 73, 81, 82, 83, 86, 90, 91, 93, 94, 99, 100, 102, 110, 120, 121, 122, 123, 132, 134, 143, 146, 151, 152, 153, 159, 160, 161, 162, 163, 175, 184, 191, 194, 197, 199, 201, 202, 203, 204, 207, 211, 212, 214, 215, 217, 221, 224, 225, 226, 233, 237, 243, 251, 252, 253, 259, 261, 263, 264, 265, 266, 268, 269, 270, 271, 272, 273, 274, 275, 276, 277, 278, 290, 297, and 299.

\*\*Ignition - fluorine: see References 141, 233, 274, 275, and 276.

Because of its strong oxidizing properties, fluorine reacts readily and often violently with most organic materials. Only the highly fluorinated hydrocarbons such as Teflon and Kel-F withstand continued service.

Cleanliness is essential for fluorine systems. If the metal surfaces in contact with fluorine or for that matter almost any strong oxidizer are contaminated with organic materials, such as traces of oil or grease, local hot spots may form which in turn may cause the violent burning of the encasing material.

It is recommended that equipment to handle fluorine be first thoroughly cleaned to remove all contaminants, e.g., organic matter, and then passivated with fluorine diluted with an inert gas.

#### FLOX\*

Mixtures of liquid fluorine and liquid oxygen (called FLOX) have received some consideration as oxidizers. Typical ratios are 40:60 and 20:80 fluorine:oxygen. The limited compatibility data that are available for FLOX are summarized in Table 7. In general, it appears that any material which performs well in liquid fluorine also performs well in FLOX.

#### Oxygen Difluoride (OF<sub>2</sub>)\*\*

Oxygen difluoride is a colorless gas and a brownish yellow liquid. It boils at -220 F. It is toxic and possesses about the same lethal characteristics as phosgene.

The limited data on the compatibility of materials with oxygen difluoride and oxygen difluoride-oxygen mixtures are summarized in Tables 8 and 9. In general, the materials behave as well as, or better than, in fluorine. No detonation has been observed in steel cylinders when filled with OF<sub>2</sub> liquid or gas and struck by 0.22-caliber long rifle bullets fired from 50 feet. (124)

#### Ozone Difluoride (O<sub>3</sub>F<sub>2</sub>)

Ozone difluoride is a viscous blood-red liquid at -297 F. At -250 F it decomposes into oxygen and oxygen difluoride.

There is little published information on O<sub>3</sub>F<sub>2</sub>. The corrosive effect of 0.05 percent O<sub>3</sub>F<sub>2</sub>-LOX on stainless steel is reported to be about the same order of magnitude as that of fluorine. (83) Compatibility data are summarized in Table 10.

#### Chlorine Trifluoride (CTF)(ClF<sub>3</sub>)\*\*\*

Like fluorine, chlorine trifluoride is among the most active chemicals known. It is a nearly colorless gas at atmospheric pressure and room temperature, but can be liquefied by the application of slight pressures. CTF reacts violently with water and many organic compounds. It attacks the respiratory tract. The threshold-limit value of toxicity for CTF in air is 0.1 ppm.

\*FLOX: see References 224, 266, and 267.

\*\*Oxygen difluoride: see References 83, 124, 224, 238, 282, and 283.

\*\*\*Chlorine trifluoride: see References 17, 23, 40, 81, 82, 106, 110, 111, 112, 113, 114, 115, 116, 117, 118, 119, 151, 158, 200, 211, 243, 244, 269, and 286.

Chlorine trifluoride can be handled in many of the common metals shown in Table 11. As with fluorine, a coating is formed on metals, which provides protection from corrosive attack. Among the metals which are resistant to CTF are copper, brass, steel, Monel, and nickel. Of these, Monel and nickel are preferred because of their resistance to hydrogen fluoride and hydrogen chloride, which are formed by reaction of CTF with water. The organic materials that are compatible are limited to Teflon and Kel-F.

Although some surface staining has occurred in tests employing impact, shock, and perforation, there has been no ignition of low-carbon steel, stainless steel, aluminum, copper, magnesium, or titanium in liquid or gaseous CTF. (114, 115)

Cleaning and passivating treatments similar to those described for metals that are used to contain fluorine must be used for CTF systems to reduce the possibility of rapid reactions.

#### Bromine Trifluoride (BTF)(BrF<sub>3</sub>)\*

Bromine trifluoride is a colorless liquid that boils at 275 F. Its vapor pressure at 70 F is 0.15 psia. It is toxic and has a threshold limit value for toxicity in the atmosphere of 0.1 ppm.

Bromine trifluoride reacts violently with many organic compounds and vigorously with water. BTF, like CTF, appears to react with some metals to form a protective coating of the metal fluoride. As shown in Table 4, this coating permits the use of nickel up to about 1300 F, copper to 750 F, and steel to 480 F. The data are not sufficiently detailed to permit a more extensive classification than that shown in Table 12.

Metals which do not form protective coatings are vigorously attacked. Examples of this type of condition are molybdenum and tungsten, either as the pure metal or in an alloy. Titanium is also attacked by BTF. Boron, silicon, columbium, and sulfur all burn in BTF (liquid).

Materials of construction, equipment design, cleaning, passivation, and general handling practice for bromine trifluoride are the same as for chlorine trifluoride.

#### Bromine Pentafluoride (BPF)(BrF<sub>5</sub>)\*\*

Bromine pentafluoride is a colorless liquid that boils at 105 F. Its vapor pressure at 70 F is 7 psia. The toxicity threshold-limit value of BrF<sub>5</sub> in the atmosphere is 0.1 ppm.

Bromine pentafluoride reacts with most of the known elements except nitrogen, oxygen, and the rare gases. Under the proper conditions, it will react with most inorganic compounds except those containing fluorine in their highest valence state. Most organic compounds react violently with BPF at room temperature and atmospheric pressure. Detailed corrosion data for metals are not available.

Recommended materials of construction for BPF are the same as those for chlorine trifluoride. The same precautions for cleaning and passivation must be followed.

\*Bromine trifluoride: see References 82, 94, 106, 151, 211, 243, and 269.

\*\*Bromine pentafluoride: see References 243 and 269.

### Iodine Pentafluoride (IPF)( $IF_5$ )\*

Iodine pentafluoride is a colorless liquid. It boils at 207 F. and has a vapor pressure of 0.4 psia at 70 F. Its toxicity threshold-limit value is 0.1 ppm.

Iodine pentafluoride is the least reactive of the halogen fluorides. Very few quantitative corrosion data for common materials of construction in IPF are available. However, it is reported that most metals are only slightly attacked by it at ordinary temperatures. The recommended materials of construction are the same as those for chlorine trifluoride.

IPF reacts violently with water. It also reacts with most organic compounds. Those rich in hydrogen will yield hydrogen fluoride and tend to ignite. Reaction with chlorine-containing compounds tends to release free iodine.

### Perchloryl Fluoride (PF)( $ClO_3F$ )\*\*

Perchloryl fluoride is a colorless gas with a sweet odor. It can be liquified at room temperature at pressures in excess of 150 psia. PF affects the respiratory tract and causes "burns" if the liquid is splashed onto the body. The toxicity threshold-limit for PF in air is 3 ppm.

Anhydrous PF is normally shipped in liquid form in steel containers. Table 13 shows other materials which resist PF quite well.

Reactions of PF with water are very slow up to temperatures of about 575 F. However, in the presence of water, PF becomes more corrosive as indicated in Table 14. Under moist conditions, Types 304, 310, and 314 stainless steels have shown relatively good resistance at room temperatures. Short-time tests also have shown the nickel alloys, Hastelloy C, titanium, and tantalum to have good resistance.

Grenade or cylinder-perforation tests resulted in detonation of titanium in liquid and gaseous perchloryl fluoride. (114,115) Titanium also ignited under impact in perchloryl fluoride but the burning was not sustained. Other metals which underwent these same tests but did not ignite were: steel, stainless steel, copper, magnesium, and aluminum. (114,115)

Teflon and Kel-F are very resistant to attack by PF. Other plastics which are suitable are unmodified phenolic resins and epoxy resins. Rubbers which are compounded with carbon black tend to be inflammable, while those with iron oxide fillers are not inflammable. The large surface areas of sponge rubbers make them inflammable. Ordinary oils, greases, and waxes should never be used with PF, however, fluorocarbon compounds are compatible. Many organic materials do not react with PF at room temperature, but if ignited will burn violently. Some inorganic materials react rapidly, e.g., mercury and "indicating" Drierite.

### Mixtures With Perchloryl Fluoride

Compatibility data for materials in 25 percent perchloryl fluoride-75 percent chlorine trifluoride

\*Iodine pentafluoride: see References 243 and 269.

\*\*Perchloryl fluoride: see References 81, 82, 83, 110, 111, 112, 113, 114, 115, 116, 117, 118, 119, 151, 211, 219, and 224.

are summarized in Table 15. With the exception of columbium, molybdenum, and titanium, most metals perform well.

Compatibility data for materials in 50 percent perchloryl fluoride-50 percent tetrafluorohydrazine are presented in Table 16. All of the usual metals of rocket construction appear compatible in both the liquid and gas at -109 F. (Tetrafluorohydrazine boils at -100 F. It can be liquified at room temperature at pressures greater than 600 psi.)

### Halogenated Hydrocarbon Propellants\*

Halogenated hydrocarbon propellants are chiefly fluorinated and chlorinated methane and ethane. Many are well known as refrigerants under the trade names of Freon, Genetron, etc. They range from gases to liquids at room temperature with boiling points of about -200 F to about +200 F.

Compatibility data for some five propellants are summarized in Tables 17 to 21. Most of the common structural materials can be used satisfactorily with the halogenated hydrocarbons. At high temperatures, some metals promote catalytic breakdown of the propellants. The general order of this tendency is: Inconel<18-8 stainless steel<nickel<copper<1340 steel<aluminum<brass<silver. (68) Aluminum alloys containing more than 2 percent magnesium and magnesium alloys are not recommended for use in halogenated hydrocarbons containing water.

Impact tests at 70 ft-lb in Propellants 113 and 114B2 have not caused ignition of aluminum, aluminum plus alumina sand, titanium, titanium plus titanium filings, or titanium plus alumina sand.

Tetrafluoroethylene and chlorotetrafluoroethylene plastics and orlon acrylic are generally suitable for use in halogenated hydrocarbon propellants. Polyvinyl alcohol resists these propellants but is sensitive to water. Phenolics, Delrin acetal resin, nylon, polyethylene and vinyls are suitable for use in many applications, but the behavior of different types may vary in different propellants, and thus should be thoroughly tested before use. Methacrylates and polystyrene are generally not suitable. No single elastomer has been found to be compatible in all halogenated hydrocarbon propellants, but a satisfactory combination can usually be found. (68)

### HYDRAZINES

#### Hydrazine ( $N_2H_4$ )\*\*

Hydrazine is a clear oily liquid with an odor similar to that of ammonia. Its vapor pressure at 80 F is 0.31 psia and it boils at 236 F. Hydrazine vapors affect the respiratory tract, nervous system, liver, and kidney, and cause "burns" when spilled on the skin. The toxicity threshold-limit value in the atmosphere is 1 ppm.

The information regarding the compatibility of various metals and nonmetals with hydrazine and

\*Halogenated hydrocarbon propellants: see References 51, 53, 68, 84, 136, 214, and 279.

\*\*Hydrazine: see References 3, 4, 7, 23, 70, 81, 82, 102, 110, 135, 145, 149, 151, 170, 171, 172, 173, 174, 183, 196, 211, and 216.

hydrazine-water mixtures is not completely consistent. These differences appear to be related to the specific application. For example, a metal may be satisfactory if air oxidation of the surface can be prevented. On the other hand, this same metal may be unacceptable for service in which prolonged exposure to air cannot be avoided.

In assessing the compatibility of a material with hydrazine, two major factors must be considered for any given exposure condition. They are:

- (1) The corrosion behavior of the material in contact with hydrazine
- (2) The effect of the material and/or corrosion products on the rate of decomposition of hydrazine.

This is particularly true for carbon and low-alloy steels, copper alloys, and molybdenum. From the corrosion standpoint, they are satisfactory. On the other hand, these metals and/or their corrosion products catalyze hydrazine decomposition at elevated temperatures. Explosions may occur. At one time, it was believed that Type 316 stainless steel (containing molybdenum) caused explosions when contacted by hydrazine at elevated temperature. However, it is now generally agreed that the hazard from misoperation in hydrazine and unsymmetrical hydrazine is no greater with 316 stainless steel than with any of the other 300 series stainless steels. (254)

Most metallic materials of construction are compatible with hydrazine. Data are summarized in Table 22.

Many plastics and rubbers are compatible with hydrazine at room temperature. Graphite and Graphitar are not suitable, since they tend to promote decomposition.

#### Monomethyl Hydrazine ( $\text{CH}_3\text{NNH}_2$ )\*

Monomethyl hydrazine is a clear liquid with the odor of ammonia. It has a vapor pressure of 1 psia at 80 F and boils at 189.5 F. Its toxic properties are somewhat similar to those of hydrazine. The recommended threshold limit is 0.5 ppm in the atmosphere.

The following materials can be used in the storage and handling of monomethyl hydrazine:

#### Metals

303 stainless steels  
304 stainless steels  
321 stainless steels  
347 stainless steels  
4130 steel  
Aluminum alloys to 160 F  
Aerobrazed-I

#### Nonmetals

Tetrafluoroethylene resins  
High-density polyethylene  
Some silicone rubbers  
Some unplasticized trichlorofluorethylene

\*Monomethyl hydrazine: see Reference 154.

Copper, lead, zinc, and alloys containing more than 0.5 wt% molybdenum are not compatible with monomethyl hydrazine.

#### Unsymmetrical Dimethyl Hydrazine (UDMH) $[(\text{CH}_3)_2\text{NNH}_2]$ \*

UDMH is a clear liquid with the odor of ammonia. Its vapor pressure at 80 F is 8.4 psia and it boils at 146 F. Its toxicity is similar to that of hydrazine but not so severe. The toxicity threshold-limit value of UDMH is 0.5 ppm.

In general, unsymmetrical dimethyl hydrazine affects materials in much the same manner as hydrazine. Of the metals, low-alloy steels, aluminum, and stainless steels are commonly used to contain UDMH. Aluminum is attacked by UDMH if water is present with the attack being in direct proportion to the amount of water. Teflon, Kel-F (unplasticized), nylon, polyethylene, and Havig 60 are among the plastic materials which are not attacked by UDMH.

Lubricants such as APS C-407, Parkerlube 5 PB, Molykote, and Peraline 12-4 may cause decomposition. On the other hand, litharge and glycerine paste, X-Pando, and Q-seal are compatible and can be used for thread compounds and other similar applications. Petroleum and silicone greases do not react, but are dissolved by the UDMH. Data for all materials are summarized in Table 23.

#### Hydrazine-Unsymmetrical Dimethyl Hydrazine Mixtures\*\*

Much of the information on properties and compatibility of 50:50  $\text{N}_2\text{H}_4$ :UDMH (Aerazine 50) is summarized in the Titan II Storable Propellant Handbook. (167,168,169) Most of the common metallic materials of construction are compatible with  $\text{N}_2\text{H}_4$ :UDMH at room temperature, including aluminum alloys, steel, stainless steel, nickel alloys, and titanium alloys. As described under "Hydrazine", Type 316 and molybdenum-containing stainless steels are no longer believed to pose an explosion hazard with 50:50  $\text{N}_2\text{H}_4$ :UDMH. (254) Of the plastic materials only some of the fluorocarbons, polyethylene, polypropylene and polyolefins are Class 1 materials. Compatibility data are summarized in Tables 24 and 25.

#### HYDROGEN\*\*\*

Liquid hydrogen boils at -423 F. Hydrogen is not toxic in the usual sense but will cause "burns" if the cold liquid contacts the skin. Hydrogen is readily ignited in air at concentrations of 4 to 74 vol%.

Liquid hydrogen and gaseous hydrogen at low temperatures are both considered to be noncorrosive. Embrittlement of metals by the low temperature of the liquid or gas is a more important factor. As can be seen in Table 26, a number of metals can be

\*Unsymmetrical dimethyl hydrazine: see References 3, 7, 23, 24, 54, 60, 61, 62, 63, 79, 81, 82, 92, 102, 110, 145, 151, 155, 156, 157, 164, 171, 172, 173, 174, 182, 196, 198, 211, 227, 228, 281, and 296.

\*\*Hydrazine-unsymmetrical dimethyl hydrazine mixtures: see References 4, 31, 47, 48, 49, 50, 165, 166, 167, 168, 169, 170, 198, 205, and 254.

\*\*\*Hydrogen: see References 34, 46, 74, 80, 81, 82, 93, 151, 181, 199, 211, and 221.

rated compatible (Class 2) with liquid hydrogen; among these are the 300 series stainless steels, Type 410 stainless steel, aluminum and most of its alloys, some nickel alloys, cobalt alloys, and molybdenum. The use of organic materials is limited because of the effect of the low temperature on their physical properties. To avoid this temperature effect, "warm joints" are used, in which the gasket material is kept at a higher temperature so that only hydrogen gas contacts a joint. Table 26 lists some of the organic materials that can be used with liquid hydrogen.

#### HYDROGEN PEROXIDE ( $H_2O_2$ )\*

Hydrogen peroxide is a colorless liquid that boils at 303 F. Its toxicity threshold-limit value in the atmosphere is 1 ppm.

When considering materials of construction for handling and containing concentrated hydrogen peroxide, both the effect of the  $H_2O_2$  on the construction material and the effect of the construction material on the  $H_2O_2$  must receive equal attention. If corrosion of the material takes place, usually the  $H_2O_2$  will also decompose, although the reverse is not true, for some materials catalytically decompose  $H_2O_2$  without much corrosive attack occurring.

One means of reducing the decomposition is to passivate the construction material before use.

#### Aluminum Passivation

An accepted passivation procedure for aluminum consists of several steps.

Step 1, thoroughly clean the metal. This step consists of degreasing with trichlorethylene, perchlorethylene, or a detergent wash, or both, depending upon the type of contaminating dirt, followed by thorough rinsing with clean water.

Step 2, treat with 5 percent nitric acid for 1 or 2 days. Rinse with tap water. Spots or areas which are not passivated can be readily identified. These spots will not have the uniform dull, velvety finish characteristic of passivated aluminum.

Step 3, reclean unpassivated areas and dig out areas containing iron or other inclusions in the aluminum.

Step 4, repeat the cleaning and nitric acid treatments until the aluminum is satisfactorily passivated.

Step 5, treat with stabilized 35 percent  $H_2O_2$  for 1 to 3 days. The passivity of the aluminum can be checked by the amount of decomposition of the 35 percent  $H_2O_2$ , by gas bubbles, and by the warming of the solution.

Step 6, rinse with distilled or deionized water.

Step 7, expose to 90 percent  $H_2O_2$ . During the first 16 to 24 hours of exposure to strong  $H_2O_2$ , equipment must be carefully watched to be sure that the  $H_2O_2$  is not decomposing.

\*Hydrogen peroxide: see References 23, 28, 35, 36, 37, 38, 41, 44, 75, 76, 77, 81, 82, 94, 101, 102, 105, 108, 110, 129, 144, 147, 148, 149, 151, 186, 187, 198, 206, 208, 210, 211, 213, 221, 229, 230, 234, 245, 246, 249, 250, 258, and 293.

Modifications of the procedure and other treatments which produce the same result may be used.

#### Stainless Steel Passivation

The passivating procedure for the 300 series stainless steels is similar to that for aluminum.

Step 1, clean with appropriate solvents and detergent solutions to remove dirt, grease, and other contamination and rinse with tap water.

Step 2, treat with 70 percent nitric acid for 4 or 5 hours and rinse with distilled water. An alternative treatment for 17-7PH steel uses a 2 percent  $Na_2Cr_2O_7 \cdot 2H_2O$ -20 percent  $HNO_3$  solution for 1/2 hour at 120-130 F.

Step 3, condition the stainless steel in  $H_2O_2$  of the concentration it will be required to handle. This operation should be observed closely to determine whether decomposition is taking place.

Step 4, if decomposition takes place, repeat the cleaning and passivating steps.

Some stainless steels may not respond to passivating treatments without prior cleaning by pickling in a 3% HF-10%  $HNO_3$  solution.

#### Pretreatment of Plastics

Plastic materials must be cleaned with the appropriate solvents or detergent solutions and rinsed. Next, they should be pre-exposed for 16 to 24 hours in the  $H_2O_2$  solution in which they will be used.

#### Discussion

Another method which effectively reduces the decomposition is the addition of sodium stannate to  $H_2O_2$  in about 2.0 ppm concentration. This treatment is effective in reducing decomposition after refilling a container with fresh stannate-free  $H_2O_2$ .

In general, the acceptability of a material for  $H_2O_2$  service is based mostly upon the amount of active oxygen that is lost by decomposition rather than upon a corrosion rate of the construction material. Accordingly, the "class" ratings used for materials for  $H_2O_2$  have been based on decomposition limits. Table 27 describes these limits in detail for four classes of materials. The table lists the material in each class immediately under the description.

Aluminum, some of its alloys, tantalum, and zirconium are the only metals included in Class 1.

Many aluminum alloys, stainless steel alloys, silicon, and tin fall into Class 2.

Class 3 contains other aluminum alloys, a variety of stainless steel alloys, Inconel X, Alloy H-075, and Refractalloys 26 and 70.

In general, the presence of copper in an aluminum alloy greatly reduces its compatibility with  $H_2O_2$ . The 1060 alloy is most widely used in 90 percent  $H_2O_2$  service; however, several other alloys are considered to be Class 1 materials.

The attack of an aluminum alloy is usually of the smooth overall type, but pitting occurs

occasionally. Pitting is usually attributed to the presence of impurities which cause local breakdown of the  $H_2O_2$ . Chloride ions also accelerate the pitting attack. The addition of small amounts of nitrate ion, such as sodium nitrate, tends to reduce the action of the chloride ion. However, the presence of 10 mg  $Cl^-$  per liter causes accelerated attack even with nitrate present. Anodizing of the metal also reduces the attack by  $H_2O_2$  with chloride, but damage of the anodic coating localizes the pitting in the damaged area. The addition of compounds to  $H_2O_2$  which change the pH in either direction from the neutral point may accelerate attack.

Galvanic coupling of aluminum to stainless steel results in increased attack on the aluminum. Chloride ions in turn increase the galvanic effect.

Many of the higher strength aluminum alloys are not compatible with  $H_2O_2$ ; therefore, one alternative is to use a strong alloy clad with a compatible grade. Of course, special attention must be given to welds, to insure complete covering of the base alloy.

Many cases of decomposition of  $H_2O_2$  in aluminum or other compatible metals is traced to soluble or suspended contamination of the  $H_2O_2$  and not to an effect of the container material.

The table shows that the 300 series stainless steels cannot be rated as Class 1, but give very good Class 2 service. The 300 series steels are used for high-pressure flowing systems and may be welded. Chloride contamination, at the 10-mg-per-liter level, does not appear to cause pitting in stainless steels. The galvanic effect in aluminum-stainless steel couples tends to protect the steel. The 17-7PH grades of stainless steel are satisfactory with  $H_2O_2$ , but the 400 series is not. A 120-grit finish on the 17-7PH steels improved their service.

Lower concentrations of  $H_2O_2$  (52 to 90 percent) require the same materials of construction as 90 percent  $H_2O_2$ . Higher strength peroxide (98 percent) is, in general, more stable than 90 percent  $H_2O_2$  in contact with metals. Aluminum alloys 1060, 5052, and 7072 are rated Class 1. Stainless steels Types 304, 316, and 347, and aluminum alloys 6061 and 356 are rated Class 2.

In the transporting of high-strength  $H_2O_2$ , 1060 aluminum has been used in tank cars and 43 aluminum or 300 series stainless steels in self-priming centrifugal pumps, while valves, fittings, and instruments are usually made of 300 series stainless steel. (208)

Of the plastics and rubbers, molded Teflon, Kel-F, and Mylar B are rated Class 1 (unrestricted use). Koroseal 700 has been used extensively as gasketing material. Many lubricants exhibit impact sensitivity in  $H_2O_2$ . The fluorinated and chlorinated lubricants appear most promising with a Class 2 rating and no impact sensitivity. Compatibility data for nonmetals in  $H_2O_2$  are presented in Table 28.

#### METHYLENE CHLORIDE ( $CH_2Cl_2$ )\*

Methylene chloride is a colorless liquid. It boils at 104 F and exerts a vapor pressure of 380 mm at 72 F. The toxicity threshold limit for  $CH_2Cl_2$  in the atmosphere is 500.

\*Methylene chloride: see References 127, 129, and 211.

Liquid methylene chloride is compatible with copper, steel, austenitic stainless steels, Hastelloy B, Hastelloy C, asbestos, and graphite. Gaseous methylene chloride is more corrosive, being compatible with Northite and Durimet 20. Compatibility data are summarized in Table 29.

#### NITRIC ACID ( $HNO_3$ )\*

The nitric acid used for propellants is usually in the concentrated form referred to as "fuming nitric acid". In general, the fuming acids contain less than 5 percent water. If the acid contains dissolved oxides of nitrogen, it is known as "red fuming nitric acid" or "RFNA". The  $NO_2$  content normally varies from 7 to 30 percent.

Nitrogen dioxide is not present in "white fuming nitric acid" or "WFNA", which contains a minimum of 97 percent  $HNO_3$ .

Hydrofluoric acid may be added to either RFNA or WFNA as a corrosion inhibitor. Listed below are the Military Specification compositions of inhibited and noninhibited acid (MIL-N-7254 C, July 19, 1956):

	<u>White Fuming Nitric Acid</u>	
	<u>Type I</u>	<u>Type I A (IWFNA)</u>
Nitric acid ( $HNO_3$ )	97.5% min.	96.8% min.
Nitrogen dioxide ( $NO_2$ )	0.0 + 0.5%	0.0 + 0.5%
Water ( $H_2O$ )	2.0% max.	2.0% max.
Hydrofluoric acid (HF)	0.0	0.6 ± 0.1%
	<u>Red Fuming Nitric Acid</u>	
	<u>Type III</u>	<u>Type III A (IRFNA)</u>
Nitric acid ( $HNO_3$ )	82.0 - 85%	81.3-84.5%
Nitrogen dioxide ( $NO_2$ )	14.0 ± 1.0%	14.0 ± 1.0%
Water ( $H_2O$ )	2.5 ± 0.5%	2.5 ± 0.5%
Hydrofluoric acid (HF)	0.0	0.6 ± 0.1%

#### Red Fuming Nitric Acid

Red fuming nitric acid is a highly corrosive material; therefore, the choice of construction materials is based upon the corrosion resistance of the material rather than on the catalytic decomposition of the acid. Aluminum and stainless steel alloys are usually used to handle RFNA. The compatibility data in Table 30 indicate that at room temperature, the corrosion rate for aluminum alloys is slightly higher than for the 300 series stainless steels. It should be noted that at 160 F, aluminum alloys are more resistant than stainless steels. Stainless steels and aluminum alloys are usually attacked in a uniform manner. However, selective attack in the heat-affected zone near welds is sometimes produced. Knife-line attack at welds may occur in aluminum alloys above 120 F. The 1060 alloy appears to be free from this attack to higher temperatures. The low-carbon grades of the 300 series stainless steels and those containing columbium or titanium are less susceptible to attack at welds than are the regular grades. Aluminum, when coupled to stainless steel, acts as a sacrificial anode to protect the steel.

\*Nitric acid: see References 18, 19, 29, 30, 32, 39, 64, 69, 81, 82, 87, 89, 93, 95, 96, 97, 102, 104, 110, 127, 128, 130, 133, 135, 150, 151, 179, 181, 185, 188, 189, 190, 192, 193, 198, 204, 211, 214, 221, 222, 223, 239, 240, 241, 242, 248, 260, 287, 298, and 300.



As little as 0.1 percent HF added to RFNA greatly reduces the corrosion rate of both aluminum and stainless steel. If this change from RFNA to IRFNA is made in stainless steel equipment and selective attack has already started, the inhibitors are not effective. The HF inhibitor reduces the selective attack at welds, permitting use at higher temperatures.

Titanium and tantalum are both resistant to RFNA; however, caution must be used with the titanium alloys. A pyrophoric reaction may occur with titanium alloys in RFNA which contains less than 1.5 to 2.0 percent  $H_2O$ . Both titanium and tantalum are attacked much more rapidly by IRFNA than by the acid without the HF addition.

A number of other alloys are compatible with red fuming nitric acid. These include cobalt alloys, Types 430 and 446 stainless steels, chromium, and for some applications, nickel and some nickel alloys. Platinum, gold, tin, and zirconium may be used. Low-alloy steels, lead, copper, and magnesium are rapidly attacked by either RFNA or IRFNA.

#### White Fuming Nitric Acid

White fuming nitric acid is similar to red fuming nitric acid with respect to compatibility with construction materials. It is not so stable as RFNA, but compatibility is largely dependent upon corrosion properties rather than on decomposition.

The corrosion behavior of metals in white fuming nitric acid is much the same as that in RFNA. The same materials are resistant and the same materials are severely attacked. However, Table 31 shows that the temperature limits are somewhat lower in the white fuming nitric acid.

#### Concentrated Nitric Acid

Table 32 has been included to show the compatibility of materials in somewhat less concentrated acids than the fuming grades. It can be seen that acids from 80 percent up to the fuming range are much more corrosive than the more concentrated ones. Stainless steels are the best materials of construction for these acids.

#### NITROGEN TETROXIDE ( $N_2O_4$ )\*

Nitrogen tetroxide is an equilibrium mixture of dinitrogen tetroxide and nitrogen dioxide ( $N_2O_4 \rightleftharpoons 2NO_2$ ). It is a heavy brown liquid that boils at 70.1 F. The liquid causes severe burns on body tissue. The toxicity threshold-limit value in the atmosphere is 5 ppm as  $NO_2$  or 2.5 ppm as  $N_2O_4$ .

Much of the information on properties and compatibility of  $N_2O_4$  is summarized in the Titan II Storable Propellant Handbook. (167,168,169)

Dry (less than 0.2 percent  $H_2O$ ) nitrogen tetroxide can readily be contained by several metals and their alloys. It is normally handled in aluminum, mild steel, cast iron, or stainless steel, although there have been reported instances of intergranular attack in welded 2014-T6 aluminum. Compatibility data are summarized in Table 33.

\*Nitrogen tetroxide: see References 3, 9, 10, 11, 12, 13, 23, 31, 47, 48, 49, 50, 59, 72, 81, 82, 88, 102, 110, 126, 131, 137, 145, 151, 165, 166, 167, 168, 169, 170, 171, 172, 173, 174, 181, 196, 198, 205, 211, 220, 221, 245, 256, 257, 287, 288, and 292.

Titanium is resistant to  $N_2O_4$  except under impact. It has been found that titanium impacts sporadically under reasonably well controlled test conditions; the ignition frequency is increased markedly by titanium filings or glass particles on the impact surface; increasing the impact-energy level increases the ignition frequency; and increasing the water content of the  $N_2O_4$  to 2 to 5 percent lowers the ignition frequency. (256,257,292) The ignitions do not spread beyond the impact area.

Moist  $N_2O_4$  is, in general, more corrosive because of the nitric acid formed. As shown in Table 34, at levels of up to 1 percent moisture in  $N_2O_4$ , most common metals are still Class 1 at room temperature. With 3.2 percent moisture in  $N_2O_4$  the corrosion resistance of steel and aluminum alloys drops off markedly even at slight increases in temperature. Data for the latter are summarized in Table 35.

Limited data on the compatibility of materials with flowing  $N_2O_4$  are presented in Table 36.

The most resistant plastic materials are polymerized fluorinated hydrocarbons such as Teflon and unplasticized Kel-F. Other plastics such as Koro-seal, Saran, polyethylene, and Tygon are suitable for short-time exposures. Vinyl plastics, in general, do not hold up well in  $N_2O_4$ .

Asbestos and graphite are satisfactory for packing materials and graphite-waterglass for thread compound.

Impact tests at 60 ft-lb in liquid  $N_2O_4$  have resulted in detonations of polydimethylsiloxane. Similar tests at 70 ft-lb did not cause detonation in polychloroprene, branched or linear polyethylene, polypropylene, polyvinylidene fluoride, polyvinylidene fluoride-hexafluoropropylene copolymer. (126)

#### OXYGEN\*

Oxygen is a light blue transparent liquid at -297.4 F. In the liquid state it can cause "burns" if spilled on the skin. Oxygen supports combustion and accidental contact should be avoided with oxidizable materials.

Of the elemental materials, oxygen is next to fluorine in reactivity. It will form compounds with all elements except the rare gases. However, the reactivity of liquid oxygen is very low compared with that of gaseous oxygen.

Liquid oxygen is considered to be noncorrosive to most metals. In particular, nickel, Monel, Inconel, copper, aluminum, the 300 series of stainless steels, brass, and silver solder are used in liquid-oxygen-handling equipment.

Several instances have been reported of violent reactions of titanium and liquid oxygen which appeared to be related to impact. The impact sensitivity of titanium in LOX has been investigated rather extensively.\*\* It appears that the ignition of titanium under impact occurs in the following sequence: (138,140)

\* Oxygen: see References 1, 5, 6, 7, 8, 21, 22, 25, 26, 27, 33, 42, 43, 45, 65, 66, 67, 71, 78, 80, 81, 82, 98, 102, 107, 125, 126, 138, 139, 140, 141, 142, 151, 176, 177, 180, 181, 195, 198, 209, 211, 218, 231, 232, 235, 236, 255, 259, 262, 284, and 285.

\*\* Ignition in LOX: see References 7, 22, 27, 42, 43, 65, 66, 71, 78, 126, 138, 139, 140, 141, 142, 176, 177, 180, 198, 218, 235, and 236.

- (1) The impact exposes fresh metal and results in some gaseous oxygen being formed at the point of impact.
- (2) The gaseous oxygen reacts with the fresh metal in an exothermic reaction.
- (3) The heat generated raises the metal temperature sufficiently to result in localized dissolution of any  $TiO_2$  film that might form.
- (4) Thus a protective oxide film does not build up and the reaction proceeds rapidly between the base metal and oxygen.

Ignition of massive titanium is observed in gaseous oxygen at liquid-oxygen temperatures at pressures of about 100 psi and above. This critical pressure limit is lowered only slightly as the temperature of the oxygen is raised to ambient temperature.

Massive aluminum exhibits ignition under severe detonation in LOX. The frequency is not so great nor is the propagation so severe as it is with titanium under the same conditions. Magnesium also ignites under detonation at shock levels higher than those for titanium but lower than those for aluminum.

Organic materials should be avoided with both liquid and gaseous oxygen because of possibilities of explosions. Currently, there is no single test or group of tests which gives a reliable compatibility rating for organic materials in liquid oxygen service. It is recommended that organic materials be avoided wherever possible and used only with caution. No completely compatible lubricants have been found. Thread antiseize sealants of graphite in chlorinated organic carrier and halogen paraffin oils with pour points as low as -100 F have been used in LOX systems. Teflon, Mylar, and certain chloroprene and Buna-N compounds have been used as static seals while Kel-F-300, Kel-F-500, Kel-F-240, Fluorothene FYTD, Fluorothene FYTS, and certain chloroprene and Buna-N compounds have been used in dynamic seals. (284,285)

Many organic and plastic materials exhibit impact sensitivity in LOX including: (71,126)

- (1) Synthetic elastomers and Thiokols
- (2) Cellulose-based papers
- (3) Silicone- and silicate-based oils and greases
- (4) Thermoplastics such as nylon and phenolics
- (5) Thermo-setting resins (phenolics, silicones, epoxies, etc.)
- (6) Petroleum-based oils and greases.

The fluorocarbon plastics are probably the best choice with respect to impact sensitivity. These, however, should not be used with aluminum. A number of other organic materials might be used, but specific conditions should be carefully studied. The list of references on ignition in LOX contains the results of many impact tests on organic materials, which can be used as a guide to selection.

## OZONE

Ozone is colorless in gaseous and liquid state. It boils at -168 F. The toxicity threshold-limit value for ozone in the atmosphere is 0.1 ppm.

There are few data on the compatibility of materials with ozone. It has been shown that 100 percent gaseous ozone can be stored up to 50 days at 5 atm pressure and dry ice temperature (-109 F) in stainless steel, aluminum, and glass with no decomposition of the ozone. (280)

## SOLID PROPELLANTS

### ANP-2639AF

There are few or no data published in the open literature on the compatibility of materials with solid propellants. Bent beam specimens of the following materials bonded to propellant ANP-2639AF and stressed to 75 percent of the 0.2 percent off-set yield strength have survived over 100 days' exposure at room temperature and 180 F. (289)

Ladish D6AC  
300 M  
Vascojet 1000  
AM-355 (longitudinal)  
PH 15-7Mo  
BI20VCA titanium  
(longitudinal and transverse)

### Nitronium Perchlorate ( $NO_2ClO_4$ )

Nitronium perchlorate is a white crystalline powder at room temperature which has a faint odor of chlorine and nitrogen oxide. Its vapor pressure is less than 0.05 mm at 68 F. It melts and decomposes at 250 to 285 F.

The following materials are reported to be compatible with nitronium perchlorate: (55)

Metals:	stainless steel (mild steel if system is dry)
Nonmetals:	glass, Teflon, unplasticized polyvinyl chloride, polyethylene
Lubricants:	Hooker Fluorolube Grease GR-54 Hooker Fluorolube Oil X-30 3M Kel-F Polymer Oil, Grade No. 1 Halocarbon Oil Series B-21.

TABLE 3. COMPATIBILITY OF MATERIALS WITH AMMONIA

Material	Temperature, F								References
	Gas				Liquid				
	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4	
<b>Metals</b>									
Aluminum	212	100	175	>175			175	175	211, 224, 287
302 Stainless Steel	75			<900					82
304 Stainless Steel	600			<900					58, 73, 109, 217, 221, 287, 288, 3, 217, 221
316 Stainless Steel	400			<900					221
347 Stainless Steel			75						109
410 Stainless Steel	600			800					58, 73, 221
430 Stainless Steel	600			<900					58, 73, 221, 221
Marbrite	75			<900					82
Durimet 20	75			<900					221
Carpenter 20	600			<900					58
Mild Steel	600			<900		75			211, 221, 287
Cast Iron	600			<900		75			211, 221, 287
St-Iron	212	75					75		211, 287
Mn-Cast Irons, low Cu	75					160			211, 287
Mn-Cast Irons, High Cu	75								All 211, 287
Nickel	500		500	<1100					75
Inconel	700		1100	<1100		75			211, 221, 287
Monel	500		500	<1100					75
Hastelloy A	600	600		<1100					58, 127, 128, 211
Hastelloy C	600	600		<1000					58, 127, 128, 211
Hastelloy D	600	600		<1000					58, 127, 128, 211
Hastelloy F		600		<1000					127, 128
Chlorimet 2-1	75								221
Nickel-Copper	75								82
Copper		75		High					Low 127, 217, 221, 287
Yellow Brass		75		High					Low 127, 221, 287
Red Brass		75		High					Low 127, 221, 287
Tin Bronze		75		High					Low 127, 221, 287
Al Bronze		75		High					Low 127, 221, 287
Si Bronze		75		High					Low 127, 221, 287
Cu-Nickel		75		High					Low 82, 127, 221, 287
Gold	212				High				211, 221
Lead		75		>160		75			211
Dow Metal C		Low							82
Dow Metal F		Low							82
Dow Metal H		Low							82
Dow Metal J		Low							82
Dow Metal M		Low							82
Platinum	212				High				211, 221
Ir-Platinum	High				High				221
Rh-Platinum	High				High				221
Silver	75								211
Ag-Cu					All				82
Titanium	175								211
Tantalum	212				High	212		High	221
Zinc	75								211
Zirconium	175								211
<b>Organic Materials</b>									
Rubber, Hard Linings				75					82
Rubber, Soft Linings				75					211
Rubber, Natural		75		Hot					82
GRS		75		Hot					82
Neoprene		75		Hot					82
Butyl Rubber		75		Hot					82
Thiokol			Cold						82
Glass Fabric and Silicone Elastomers*	Hot								82
Silicone Greases	Hot								82
Neveg 41 Epon	212								211
Silicone Elastomer		75							82
Silicone Resins									82
Teflon	Hot								82
Corb									82
Vinyl Copolymers		Hot				Hot			82
Phenolics		Hot				Hot			82
Purans		Hot				Hot			82
Polyethylene		Hot				Hot			82
Kel-F		Hot				Hot			82
Vinylidene Chloride			Cold				Cold		82
Sulfur Cement			Cold				Cold		82
Aluminum Composition			Cold				Cold		82
Polystyrene				75					211
Polyesters				75					211
Phenol Formaldehyde				75					211
<b>Nonmetals</b>									
Glass	212								211
Stoneware	212								211
Silicate	>2000								82
Carbon	>2000								82
Graphite	>2000								211

\* Glass Fabric and Silicone Rubber

TABLE 4. COMPATIBILITY OF MATERIALS WITH HI-Cal-3 (82)

Material	Remarks
<b>Class 1</b>	
Mild steel	Below 120 F, probably higher
40/50 carbon steel	Below 120 F, probably higher
Hot-rolled primer steel	Below 120 F, probably higher
Type 304 stainless steel	Below 120 F, probably higher
Beryllium copper	Below 120 F, probably higher
Cupro-nickel	Below 120 F, probably higher
Naval brass	Below 120 F, probably higher
Phosphor-bronze	Below 120 F, probably higher
Phosphorized copper	Below 120 F, probably higher
Nickel	Below 120 F, probably higher
Inconel	Below 120 F, probably higher
Monel	Below 120 F, probably higher
Niobel	Below 120 F, probably higher
Incoloy	Below 120 F, probably higher
Aluminum	Below 120 F, probably higher
Lead	Class 2 at 120 F
Titanium	Below 120 F, probably higher
Tantalum	Below 120 F, probably higher
<b>Class 2</b>	
<b>Organics</b>	
Bakelite	No change at 120 F
Easton PVC plastic pipe	No change at 120 F
Epoxy resin	No change at 120 F
Graphite bearing	No change at 120 F
Kel-F 300	No change at 120 F
Kel-F 500	No change at 120 F
Teflon	No change at 120 F
Hycar 1001-520-39-5-4	No change at 120 F
Viton 4411A-58	No change at 120 F
Hycar 1000x88-520-39-20-3	No change at 120 F
Hycar 1001-520-37-83-1	No change at 120 F
IF4 Fluororubber	No change at 120 F
LS-53 Fluorosilicone rubber	Slightly less resilient at 120 F
Hycar 1001-520-37-83-5	No change at 120 F
X-Pando pipe dope	No change at 120 F
Nylon Zytel 101-KC-10	No change at 120 F
<b>Class 3</b>	
<b>Organics</b>	
Hycar 1001-520-39-5-2	120 F, Slightly stiffened
Polyethylene tubing	120 F, Turns yellow
Garlock 8748 (Buna-W binder)	120 F, Stiffened
Garlock 7021 (GRS - high sulfur binder)	120 F, Stiffened and roughened
Garlock 900 (GRS binder)	120 F, Stiffened
African Blue Asbestos packing	120 F, Darkened
Teflon asbestos packing	120 F, Weight gain
<b>Class 4</b>	
<b>Organics</b>	
Hycar 1001-520-39-5-3	120 F, Blistered
Compressed asbestos gasket	120 F, Fibers loosened
Garlock 7228 (neoprene binder)	120 F, Blistered
Hycar 1042-520-24-144-1	120 F, Brittle, crazed
Tygon tubing	120 F, Hardened
Garlock 7705 (GRS - blue asbestos)	120 F, Stiffened
Plexiglas	120 F, Became soft and sticky
Fairprene 5051 (neoprene on duck)	77 F, Stiffened
Fairprene 5039 (neoprene on nylon)	77 F, Became brittle
Natural rubber	120 F, Softened, easily torn
National 846, O-ring	120 F, Stiffened
Neoprene	120 F, Softened
Hycar 1001-520-39-5-1	120 F, Stiffened
Hycar 1001-520-39-5-3	120 F, Stiffened
Silicone rubber	120 F, Deteriorated to a powder

TABLE 5. COMPATIBILITY OF MATERIALS WITH PHTHALONITRILE (204)

Material	Temperature, F								References
	Gas				Liquid				
	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4	
Metals									
Aluminum, 2024 T-3					75				
Aluminum, 3003 H-14					75				
Aluminum, 5052 H-14					75				
Aluminum, 6061 T7-6					75				
Aluminum, 7075 T-6					75				
Aluminum, 356 T-6					75				
Aluminum, Chromated					75				
Cadmium-Plated Steel					75				
Cadmium-Coated Aluminum					75				
Copper					75				
Brass					75				
Iron					75				
Steel					75				
302 Stainless Steel					75				
304 Stainless Steel					75				
321 Stainless Steel					75				
347 Stainless Steel					75				
Magnesium Alloy, Fed QQ					75				
Nickel					75				
Nickel Alloy, X125H					75				
Titanium, C-110H					75				
Titanium, C-130H					75				
Nonmetals									
Asbestos, Graphite					75				
Asbestos, Hercules No. 571 K					75				
Pure Carbons					75				
Graphite No. 39					75				
Nylon					75				
Nylon A					75				
Nylon B, No. 500C					75				
Polycarbonate and Glass Cloth					75				
Polystyrene					75				
Fluorinated Rubber					75				
Foamglass									75
Dow Corning R-7502 Foam									75
Dow Corning R-7003 Foam									75
Mopco F 10 Foam									75
Mopco B 40									75
Natural Rubber									75
Nitrile Rubber or Nylon									75
Dow Corning 9483 Rubber									75
Low-Density Plastic 25G									75
150-101									75
Boron									75
Nylon									75
Nylon									75
Nylon									75
Rubetek C-20/27N									75
Rubetek A-103J									75
Pierreflex, No. XSR									75
Pierreflex, No. 51P									75
Dow Corning Silastic									75
No. 80-24-48C									75
Geacica Silastic No. 24C									75
Lubricants									
Rockwell Nordtron Lube					75				
No. 92					75				
Molybdenum Disulfide					75				
Rockwell Nordtron Lube									75
No. 833									75
Rockwell Nordtron Lube									75
No. 8-21									75
Rockwell Nordtron Lube									75
No. 840									75
Rockwell Nordtron Lube									75
No. 356									75
Rockwell Nordtron Lube									75
No. 822-S									75
Rockwell Nordtron Lube									75
No. 8-20									75
Rockwell Nordtron Lube									75
No. 942-S									75
Water-Based Lubricants									75

TABLE 6. COMPATIBILITY OF MATERIALS WITH FLUORINE

Material	Temperature, F								References
	Gas				Liquid				
	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4	
Metals									
Aluminum 1100	500	600	800	<673	-320			-310	7,73,132, 143,153,159, 207,211,274
Aluminum 2017						-320		-310	143
Aluminum 3003	700	>1000	1000	>1000		-320		-310	120,143
Aluminum 5052	700	700	1000	>1000		-320		-310	120,274
Aluminum 5154	700	>1000	1000	>1000		-320		-310	143,274
Aluminum 6061						-320		-310	153
Aluminum 7075						-320		-310	274
Beryllium				75					269
304 Stainless Steel	400	400	500	>900	-320	-320	-310		58,211,274
304 L Stainless Steel	270		400	>900					143
309 Stainless Steel	570		750						7
309 Co Stainless Steel	500		750	>900					207
316 Stainless Steel	570		660	750					7,207
321 Stainless Steel	400				-320				58,211,274
347 Stainless Steel	390	570	750		-320		-310		120,274
410 Stainless Steel				75	-320		-310		58,153,211
430 Stainless Steel					-320				274
430 Stainless Steel	400	400	500	600	-320				7,207,211
Carpenter 20	400				-320				58
PH15-NiFe					-320				153,274
AM-350-C					-320				274
AM-350-CK					-320				274
AM-350-D					-320				274
AM-350-DE					-320				274
Cast Iron				75					211
Armco Iron	750		500	>900					7,159,163,207
Stellite Iron				75					211
Iron (S.304S1)		200	400	>900					159,160
Iron (S.74S1)		100	300	>900					159,160
Sheet Steel	750	840		930					7
SAE 1010	100	200	400	>900			-310		207
SAE 1015	300	400	570	>900					7,159,163, 207,211
SAE 1020	390			930					7,159
SAE 1130	750		840	930					7,159
Carbon Steel	750		840	930					82
Mild Steel		570							7
Al-Nickel	1200	900	1000	>1300	-320		-310		7,73,120,143 153,159,160, 211,274
Ni-Nickel		1200	1000	>1300			-310		82
Nickel (Low Carbon)	1100	1200	1000	>1300					82
Nickel (Electrolytic)	1100	1200	1000	>1300					82
Dynalene	1100	1200	1000	>1300					82
Monel	1200	1200	1000	>1300	-320	-320	-310		7,73,143, 153,159,160, 207,211,274
Cast Monel	400	800	1000	>1300			-310		82
Al-Monel	600	800	1200	>1300					82
Inconel	930	1170	500	>900			-310		7,207,211
Hastelloy B		75	212						58,127,128, 211
Hastelloy C	95		212						58,127,128, 211
Hastelloy D			212						58,211
Copper	200	400	800	>900	-320		-310		7,153,159, 207,211,274
Deoxidized Copper		230	1290						7
Copper ETP	200	400	800	>900					82
Brass (70-30)	200								159,160
Brass (80-20)	200								73,211
Brass (24-76)	200	400	600	>900			-310		82
Brass (Low-leaded)			500	>900					120
Brass (Yellow)					-320				274
Brass (Cartridge)					-320				274
Brass (Castings)									274
Brass	200	400	700				-320		73,159,160, 211
Copper-Nickel	200	400		>900	-320				155
Copper-20% Nickel					-320				274
Copper-30% Nickel					-320				274
Everdur 1010					-320				274
Chromium Plate	400								159,160
Lead			100	>100					211
Magnesium		500	700						82
Magnesium M1A	653	<1000							143
Magnesium AZ1A-T6	691	<1000							143
Magnesium ZK21G-T6		400					-310		82
Magnesium AZ91G-T6	<1000								143
Magnesium HE-31					-320				274
Magnesium HE-31A-M24			1100				-310		82
Magnesium HE-31 (Coated with Dow-17)							-320		274
Magnesium HE-31					-320				153,274
Magnesium AZ-31					-320				274
Magnesium AZ-31 (Coated with Dow-17)					-320				274
Magnesium MA (1.20Sn)		200							159,160
Magnesium PS-1A		200							159,160
Magnesium J-1H	140								159,160
Magnesium H-1A		600	1000				-310		82
Magnesium Dow Metal G	570	600							7,207



TABLE 9. COMPATIBILITY OF MATERIALS WITH 70 PERCENT  $O_2$ -30 PERCENT  $O_2$  (1 ATMOSPHERE) (124)

Material	Temperature, F								Reference
	Gas				Liquid				
	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4	
<u>Metals</u>									
Aluminum, 2024-T4	212								
Aluminum, 2017-SF	212								
Copper	212								
Brass	212								
Mild Steel	212								
303 Stainless Steel	212								
304 Stainless Steel	212								
316 Stainless Steel	212								
318 Stainless Steel	212								
Wrought	212								
Lead	212								
Magnesium, Dowmet: FS14	212								
Magnesium, Dowmet: FS1A	212								
Magnesium, Dowmet: MA	212								
Nickel	212								
Ni Monel	212								
P Monel	212								
304 Monel	212								
5 Monel	212								
8 Monel	212								
CR Monel	212								
Inconel	212								
Hastelloy D	212								
Platinum	212								
<u>Nonmetals</u>									
Teflon	212								
Ref: 1			212						

TABLE 10. COMPATIBILITY OF MATERIALS IN OZONE DIFLUORIDE-LIQUID OXYGEN (1.00% OF<sub>2</sub> - BALANCE LOX) (83)

Material:	Temperature, °F								References
	Gas				Liquid				
	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4	
304 Stainless Steel	-297				-297				
316 Stainless Steel	-297				-297				
347 Stainless Steel	-297				-297				
416 Stainless Steel		-297				-297			
Al-19-Mg	-297					-297			
Al-506	-297				-297				

TABLE 11. COMPATIBILITY OF MATERIALS WITH CHLORINE TRIFLUORIDE (CITF)

Material	Temperature, F								References
	Gas				Liquid				
	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4	
<b>Metals</b>									
Aluminum, 1060					85				121, 125
Aluminum, 1100	85		644		85				121, 125, 127, 128, 211, 215, 217
Aluminum, 1100, Welded	85				85				127
Aluminum, 2014, Welded	85				85				127
Aluminum, 2024					85				121, 125
Aluminum, 3003					85				121, 125
Aluminum, 5052					85				121, 125
Aluminum, 6061, Welded	85				85				127
Aluminum, 7079					85				121, 125
Columbium								105	121, 125
Copper, ETP	85		500		85				121, 125, 127
Copper, DHP					85				121, 125
Beryllium-Copper, 2%					85				121, 125
Phosphor Bronze, 95					85				121, 125
Aluminum Bronze, 95	85				85				121, 125, 211
Yellow Brass	85				85				121, 125
Red Brass					85				121, 125, 211
tin Bronze	75				85				121, 125
1010 Steel	85		645		85				121, 125, 127, 211
1015 Steel Coated with Fosbond 40					85				121, 125
1015 Steel Coated with Fosbond 21	85				85				121, 125
304 Stainless Steel					85				121, 125
304 Stainless Steel					85				121, 125, 211
316 Stainless Steel	85				85				121, 125
347 Stainless Steel					85				121, 125
AM-35C									82
A-286									82
Carpenter 20					85				82, 121, 125
PH15-7Mo (H 900)					85				127
PH15-7Mo (H 1050)					85				127
410 Stainless Steel (Welded)	85				85				127
304 Stainless Steel (Welded)	85				85				127
AM-35C (Welded)	85				85				127
PH15-7Mo (Welded)	85				85				127

- TABLE 11. (Continued)

Material	Temperature, F								References
	Gas				Liquid				
	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4	
Lead		74							17
Magnesium, AZ-31B	95				85				111,115
Magnesium, HK-21A					85				111,115
Magnesium, HK-31A					85				111,115
Aluminum	85				85				111,115,116, 211
Monel	85				85				111,115,116, 211
Inconel					85				111,115
Inconel X					85		7		81
Incoaloy					85		7		111,115
Hastelloy X						7	7		82
Werk 41									82
Super-Nickel, 305					85				111,115
Nickel, 200 (Welded)	85				85				111,115
Monel 400 (Welded)					85				111,115
Nickel-Silver, 18K (Alloy A)					85				111,115
Thorium		85							82
Titanium, 100A				75					87,111,115
Titanium, G-102AV									111,115
Titanium, A-110AT									111,115
Zirconium				600					82
Zirconium				600					87
<b>Nonmetals</b>									
Glass	75								211
Pyrex Glass		75							82
Stoneware	75								211
Neoprene Rubber	75								211
Carbon, Korbate #154									111,115
Graphite, Korbate #254									111,115
Graphite Graphitar 59									111,115
Graphitar 6"						75			111,115
Graphite A-1000									111,115
Teflon	75				85				111,111,115, 211
Kel-F		75				85			111,111,115
Polymethylsiloxane Fluoride (MC-2525)					85			130	111,115

- a) Epoxy filler.
- b) Phenolic filler.

TABLE 12. COMPATIBILITY OF MATERIALS WITH BROMINE TRIFLUORIDE (BrF<sub>3</sub>)[illegible]











TABLE 24. COMPATIBILITY OF MATERIALS WITH 50:50 HYDRAZINE-  
UNIONIZED DIMETHYL HYDRAZINE (50:50) (UHM)

Material	Temperature, F								References
	Gas				Liquid				
	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4	
Metals									
Aluminum, 1100					60				21, 165, 169
Aluminum, 1100-O					60				60
Aluminum, 2014-T4					60				20, 169
Aluminum, 2014-T6	160				160				20, 165
Aluminum, 2014-T6 (Welded)	160	60			160	60			169
Aluminum, 2014-T6 (Spot Welded)							60		169
Aluminum, 2014-T6 (Extrusion)	160				160				20, 169
Aluminum, 2014-T6 (Extrusion Stressed to 30,000 psi)					60				169
Aluminum, 2014-T6 (Welded and Stressed to 30,000 psi)					60				169
Aluminum, 2014-T6 (Hard Anodize)					60				20, 169
Aluminum, 2014-T6 (H2SO4 Anodize)	160				160				169
Aluminum, 2014-T6 (Iridite)					60				20, 169
Aluminum, 2014-T6 (Alodine)		160				160			169
Aluminum, 2014-T6 (Fluoride)					60				169
Aluminum, 2024-T6	160				160				20, 165, 169
Aluminum, 2219-T81					60				169
Aluminum, 2219-T81 (Welded)	130				130				169
Aluminum, 3003-H14	160				160				169
Aluminum, 3003-H36					160				169
Aluminum, 3003-H19					160				169
Aluminum, 3254-F	160				160				20, 169
Aluminum, 3456-H14					60				169
Aluminum, 3456-H36					60				169
Aluminum, 3456-H19					60				169
Aluminum, 3456-H321	160				160				169
Aluminum, 3456-H321 (Stressed 30,000 psi)					60				169
Aluminum, 3456-H321 (Welded)	60				60				169
Aluminum, 3456-H321 (Welded and Stressed to 30,000 psi)	60				60				169
Aluminum, 6061-T6	160				160				20, 165, 169
Aluminum, 6061-T6 (H2SO4 Anodize)	160				160				169
Aluminum, 6061-T6 (Iridite)	160				160				169
Aluminum, 6061-T6 (Alodine)	160				160				169
Aluminum, 6061-T6 (Fluoride)	160				160				169
Aluminum, 7075-T6	160				160				165, 169
Aluminum, 7075-T6 (Stressed to 80% of Yield)					160				169
Aluminum, 7075-T6	160				160				165, 169
Aluminum, 7075-T6	160				160				169
Beryllia 25	160				160				165, 169
Cadmium Plate					60				169
Chromium Plate					160				169
Stellite 20	160				160				169
Stellite 6K	160				160				169
Stellite 21	160				160				169
Copper Plate					60				169
718 Filler Bronze of 6061-T6 Al					160				169
AMS 4775 Microbrass of 347 Stainless Steel					160				169
C-62 Bronze (W-Mn-Co) of 347 Stainless Steel					160				169
Gold Plate	160				160				169
1020 Steel					60				20, 169
4130 Steel	160				160				20, 169
303 Stainless Steel	160				160				20, 169
304 L Stainless Steel	160				160				20, 169
316 Stainless Steel (Welded)	160				160				169
321 Stainless Steel	160				160				169
347 Stainless Steel (Welded)	160				160				169
347 Stainless Steel (Welded)	160				160				20, 165, 169
410 Stainless Steel (Welded)	160				160				169
410 Stainless Steel (Welded)	160				160				169
440C Stainless Steel (Welded)	160				160				169
A-286	160				160				165, 169
PH13-7Mo (Cond. A)	160				160				169
17-4PH	160				160				169
17-7PH (Cond. A)	160				160				169
90-10Ti (Cond. A)	160				160				169
90-10Ti (Cond. B)	160				160				169

TABLE 24. (Continued)

Material	Temperature, F								References	
	Gas				Liquid					
	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4		
Magnesium, AZ31									160	169
Magnesium, AZ31A-T6									60	31, 30, 165, 169
Microsteel 100-1 on AM100A, Mg				160					160	169
Microsteel 100-100 on AZ31C, Mg				160					160	169
Nickel	160				160					165, 169
Ni Span C					60					165, 169
Silver					60					50, 169
Nickel Electroplate					60					169
Nickel, Electroless Plate	160				160					169
Silver Solder					60					50, 165, 169
Easy Pin 45						100				169
Easy Pin Silver Brass of 347 Stainless Steel					100					169
Silver Plate					60					169
Tin Plate					60					169
Pure Tin Solder of 303 Stainless Steel					160					169
Titanium, B1000A	160				160					50, 165, 169
Titanium, A100A-T	160				160					169
Titanium, C100A-T	160				160					165, 169
Titanium Carbide (Nickel Binder)	160				160					169
Tungsten Carbide	160				160					169
Zinc Plate									50	169
Plastics										
Teflon (TFE)					80	15				50, 92, 165, 169
Teflon filled with Graphite					60					50, 169
Teflon filled with MoS <sub>2</sub>					60					50, 169
Teflon filled with Asbestos					60					50, 169
Armalon 7700 with Teflon Fibers								50		50, 165, 169
Armalon 7700 with Teflon Fibers								50		50, 165, 169
Fluorobutene filled with Asbestos					60					169
TFE-Pelt 7500					60					50, 165
Fluoropropene					60					50, 165, 169
Teflon (FEP)					60				80	169
Kel-F 300 Unplasticized							60			165, 169
Kel-F 300 Annealed							60			50, 169
Kel-F 300 (10% Glass Filled)					75					92
Kel-F 300 (Unfilled)					75					92
Kel-F 300					75					92
Low-Density polyethylene					60					50, 169
High-Density Polyethylene							60		160	169
Martex 50 Polyethylene					60					50, 169
Polyethylene 7028							80			92
Polyolefin, White					160					169
Insulation								160		169
Polyolefin, Black					160					169
Polypropylene					160					50, 169
Zytel 31								60		169
Zytel 63								60		169
Zytel 101								60		169
Nylar								60		50, 169, 165
Nylar A							75			92
Silicone-Glass Laminate								60		50, 169
Phenolic-Glass Laminate								60		50, 165, 169
Epoxy-Glass Laminate								60		50, 165, 169
Polyester-Glass Laminate								60		169
Saran								80		50, 169
Dacron								60		50, 169
Lexan								60		50, 169
Tedlar								60		169
Kynar								60		169
Plexiglas 28 39								60		50, 169
Plexiglas 11								80		169
Opalon 1219								60		169
Opalon 1220								60		169
Opalon 1444								60		169
Opalon 81222								60		169
Aeroplaste								160		169
Tygon								60		169
Rigid PVC								60		169
Epon 828								60		169
Epon 828 (with PMDA)								60		169
BC 1469 Epoxy								60		169
Hypalon 20								60		169
Phenolic Asbestos								60		169
Si-20-80								60		169
Silicone A-70X								60		169
Normco X3168								60		169
P-4019								60		169
BC-1000								60		169
N-Film								75		169
Dupon 30 (Glass Filled)								75		92
Dupon 30 (Unfilled)								75		92



TABLE 25. COMPATIBILITY OF MATERIALS WITH 50:50 HYDRAZINE (UNSYMMETRICAL DIMETHYL HYDRAZINE) PLUS UP TO 3% WATER (50)

Material	Temperature, F								References
	Gas				Liquid				
	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4	
<b>Metals</b>									
Aluminum, 1100-O					>60				
Aluminum, 5656H321					>60				
Aluminum, 7075-T6					>60				
Aluminum, 356-T6	>60								
Ni-Span-C					>60				
Titanium, 5A1-4V					>60				

TABLE 26. COMPATIBILITY OF MATERIALS WITH LIQUID HYDROGEN (82,211,221)

Class 1 or 2 Materials		Class 4 Materials
Aluminum, 1100	Inconel	Aluminum, 40 E
Aluminum, 5052	Low-Carbon Steel	Magnesium
Aluminum, 4043	High-Nickel Steels	Lead
Aluminum, 2024-T	Titanium	Zinc
Aluminum, 1100-T	Nitrile Rubber	Iron
300 Series Stainless Steel	Silicone Rubber	High-Carbon Steel
410 Stainless Steel	Teflon	Natural Rubber
Haynes 21	Gerrick Packing	Neoprene Rubber
Molybdenum	Barexite	Cork
Nickel	Micarta	Wood
Monel	Lucite	Polyvinyl Chloride
	Graphite	Saran
		Polyvinyl Alcohol

TABLE 27. COMPATIBILITY OF METALS WITH 90 PERCENT HYDROGEN PEROXIDE

Aluminum Alloys	Stainless Steel Alloys	Nickel Alloys	Cobalt Alloys	Other Alloys	Other Pure Metals	Reference
<b>Class 1 Metals</b>						
Maximum per cent of active oxygen loss (AOL) by H <sub>2</sub> O <sub>2</sub> in 1 week: at 30 C (76 F) - 0.2%, at 66 C (150 F) - 5.0%						
Minimum stability of H <sub>2</sub> O <sub>2</sub> after test - 95% stable in 24-hour test at 212 F in glass						
No other effect on H <sub>2</sub> O <sub>2</sub> or metal						
1060	None	None	None	None	Tantalum	35,36,37, 38,149,210
1100					Zirconium	35,36,132, 149,210
1160						35,210
1260						35,210
7072						35,36,37, 38,210
B-356						35,36,38, 210
<b>Class 2 Metals</b>						
Maximum per cent of active oxygen loss (AOL) by H <sub>2</sub> O <sub>2</sub> in 1 week: at 30 C (76 F) - 6.0%, at 66 C (150 F) - 80%						
Minimum stability of H <sub>2</sub> O <sub>2</sub> after test - 90% stable in 24-hour test at 212 F in glass						
No other effect on H <sub>2</sub> O <sub>2</sub> ; slight bronzing of the metal allowable but no corrosion						
3003	Type 202	None	None	None	Silicon	35,36,210
4043	Type 302				Tin (cp)	35,36,149, 210
5052	Type 304					35,36,37, 38,149,210
5054	Type 304 ELC					35,210
5056	Type 309					35,210
5652-0	Type 310					35,36,210
5254-0	Type 316					35,36,37, 38,210
6061	Type 316 ELC					35,36,37, 38,210
3063	Type 317					35,210
6363	Type 318					35,36,210
150	Type 321					35,210
214E	Type 322					35,36,37, 38,210
214F	Type 347					35,36,210
356F	17-7PH, 37-45 R <sub>c</sub>					35,36,37, 210
5052	17-7PH, 45 R <sub>c</sub>					35,210
(anodized)	(buffed)					
6061	17-7PH					35,210
(anodized)	(unhardened)					
6061	Hasco-O-Seven					35,210
(HNO <sub>3</sub> Pass.)						
6061	Malin-Wilabrite					35,36,210
(detergent wash)						
6061						35
(WFNA Pass.)						

Table 27. (Continued)

Aluminum Alloys	Stainless Steel Alloys	Nickel Alloys	Cobalt Alloys	Other Alloys	Other Pure Metals	Reference
<u>Class 3 Metals</u>						
Maximum per cent of active oxygen loss (AOL) by H <sub>2</sub> O <sub>2</sub> in 1 week: at 30 C (76 F) - 11.0%, at 66 C (150 F) - 100% in 24 hours Minimum stability of H <sub>2</sub> O <sub>2</sub> after test - 15% after 30 C (76 F) test Bronzing and staining, but no rusting or other corrosion products; slight attack may be allowed.						
2024	329	Inconel X	None	H-975	None	35, 37, 38, 210
B214F	AM 350			Refractalloy 26		35, 210
A360	17-7PH, 45 Rc			Refractalloy 70		35, 210
6061	17-7PH, 45 Rc					35
(anodized)	(electropolished)					
2017	17-4PH					35
(anodized)						
2024	Carpenter 20					35, 36, 210
(anodized)	Durimet 20					35, 210
6061	AM 355					35, 210
(hard coat)	19-9DL					35, 210
	16-25-6					35
	Preloy Type					35, 210
	302 porous wire					
	Worthite					35, 210
<u>Class 4 Metals</u>						
Maximum per cent of active oxygen loss (AOL) by H <sub>2</sub> O <sub>2</sub> in 1 week - not specified Minimum stability of H <sub>2</sub> O <sub>2</sub> after test - not specified. Pitting or corroding during or after test						
2014	AM 350	Nickel	Cobalt	Mild steel	Copper	35, 36, 149, 210
2017	410	Inconel	Haynes 25	Be-bronze	Zinc	35, 36, 149, 210
7075	416	Monel	Star "J"	Chromaloy H-3	Tungsten	35, 149, 206, 210
40E	420	Hastelloy	Haynes 3	Duriron	Titanium	35, 36, 210
218	430	"B"	Haynes 6	(cast)	Sodium	35, 36, 210
355F	431	Hastelloy	Haynes 12	Fanweld "O"	Magnesium	35, 38, 149, 210
		"C"				
A750	440		S-588	Ni-Resist	Beryllium	35, 210
B750	443	Hastelloy	S-590	Tantung	Cadmium	35, 149, 210
2024	446	"D"		Dow metal JIA	Chromium	35, 149, 210
(anodized)	Rigimesh J	Be-nickel		Dow metal MA	Gold	35, 210
	SS porous	Illum "G"		Kennametal	Iron	35, 36, 210
	wire	Chlorimet		K-138	Lead	35, 36, 210
	300 series SS			Kennametal	Manganese	35, 210
	powder compact			K-3H	Mercury	35, 210
	Type 302			Kennametal	Molybdenum	35, 94, 149, 210
	powder compact			K-501		
					Platinum	35, 210
	Type 316			Kennametal	Silver	35, 36, 210
	powder compact			K-M		
	Type 302B			Multimet N-155		35, 210
	powder compact					
	Type 316 + Cb					35
	powder compact					
	Utiloy 3					35, 210
	Utiloy 20					35, 210
	Utiloy H					35, 210
	Utiloy NH					35, 210
	Elgiloy					35, 210

TABLE 28. COMPATIBILITY OF NONMETALS WITH 90 PERCENT HYDROGEN PEROXIDE AT 150 F<sup>(35)</sup>

Material	Classification
<u>Plastics - Polyethylene and Halogenated Polyethylene Type</u>	
Fluoroflex I-TP1001	1
Fluoroflex I-TP1000 (black)	2
Fluran B-4100	3
Halgene	2
Hypalon S-2	4
Hypalon Gasket	4
Hypalon V-54-B (gray)	4
Hypalon V-56-A (gray)	4
Hypalon V-163-4 (black)	4
Hypalon "O" Ring (GRC 90-5)	3
Irrathene 101 (irradiated polyethylene)	2
Kel-F (unplasticized)	1
Kel-F 800 (Lot 5649)	1
Kel-F 820 (G4028)	2
Kel-F 3700 gum	3
50% Kel-F 3700-50% Kel F 800	2
Kel-F 5500 (unpigmented)	2
Kel-F 5500 gum	3
Kel-F 5500-121	2
Kel-F 5500-61	2
50% Kel-F 5500-50% Kel-F 800	1
75% Kel-F 5500-25% Kel-F 800	1
Kel-F O-Ring (CPD. 7761-70)	2
Polyethylene	2
Rulon (Teflon Base)	2
Teflon (white)	1
Viton A (411A4) (black)	3
<u>Plastics - Polyvinylchloride and Copolymers</u>	
Alanol tubing	3
Boltron 6200 (gray)	2
Geon 118	4
Geon 404 (yellow)	3
Koroseal 116	3
Koroseal 117 (molded)	3
Koroseal 700	2
Lucoflex (translucent)	3
Lucoflex (white)	3
Marvinol 218-200	4
Marvinol 218-201	4
Marvinol NG-3005	4
Marvinol NR-6010	4
Saran	2
Saran Rubber Q-187	4
Transflex Tubing	4
Tygon B-20	3
Tygon B-32	3
Tygon B-63	3
Tygon B-71	3
Tygon B-72	3
Tygon B-136	3
Tygon S-22-1	4
Tygon TL-103	4
Tygon 2807	4
Tygon 3400	4
Tygon 3603	4
Tygon 3604A	2(a)
Tygon 3604B	2(a)
Vinyl 79139	2
Vinylite VG 1914	2
Vinylite VU 1940	2
Vinylite VS 1310	3
Vinylite VU 1900	3
Vinylite UE 1907	3
Vinylite VU 1920	3

TABLE 28. (Continued)

Material	Classification
<u>Plastics - Polyvinylchloride and Copolymers</u>	
Vinylite VU 1930	3
Vinylite VU 1940	3
<u>Plastics - Silicone Rubber Compounds</u>	
Fluorosilicone LS-53	2
GE 407B-217-1	4
GE 1240	2
GE 81223	2
GE 12601	4
GE 12602	3
GE 12650 (unpigmented)	2
GE 12650 (pigmented red)	3
GE 12670	4
GE 12670 (pigmented brown)	4
GE 15060 (pigmented)	3
GE 15080	3
GE X-7181	3
Parkone White 467-1 O-ring	4
SE 450 (unpigmented)	2
Silastic 152	3
Silastic 160	3
Silastic 160 O-ring	4
Silastic 161	3
Silastic 181	3
Silastic 240	2
Silastic 250	4
Silastic 261	3
Silastic 675	3
Silastic 6-128	2
Silastic 7-180	3
Silastic 9711	2
Silastic HR-9711	2
Silastic 9711 welded with S-2200	2
Silastic S-2000-4-480	2
Silicone 407-B-217-1	3
Silicone 407-B-437-1	2
Silicone HT 656	3
Silicone SR 5550	2
Silicone SR 5570	2
SR 5550	2
SR 5570	2
X-7181	3
Silicone Y-1749	2
<u>Rubbers and Plastics - General</u>	
Acrylon Rubber BA-12	4
Acrylon EA-5	4
Adiprene C	4
Bisilon No. 50	2
Buna N	4
Butyl Rubber A 3405	4
Butyl Rubber SR-384	4
Cyclocac (natural color)	4
Garlock No. 5681 (Teflon-impregnated asbestos)	4
Hycar PA 478-1-1 (black)	4
Haveg 41 (Asbestos filled phenolic)	4
Haveg 60 (phenolic)	4
Hysol 4-77C (clear)	4
Hysol 4-77D (amber)	4
Hysol 4-77E	4
Hysol 4-77F	4
Hysol 4-78A (white)	4
Hysol 4-78B (brown)	4
Hysol 4-78C (amber)	4
Hysol 4-78D (amber)	4
Hysol 6000 B (amber)	4

TABLE 28. (Continued)

Material	Classification
<u>Rubbers and Plastics - General (continued)</u>	
Kralite	3
Melmac No. 1077	4
Mylar A	1
Mylar B	1
Neoprene Pure Gum	4
Neoprene SR 365-B	4
Nylon	4
Phenol-Formaldehyde	4
Plexiglas	4
Polystyrene (Polyflex)	2
Thiokol EC-801-LP2	4
Thiokol 3000 FA	4
Thiokol 3000 ST	4
Thiokol 1620 AH	4
<u>Plastics - Laminates, Diaphragms</u>	
<u>Materials and Adhesives</u>	
Chemeloc MI-411 (Teflon Fiberglass)	2(b)
Duroid 5600 (fiber-reinforced Teflon)	3
Fairprene PS57-167 (Viton A, 116 Glass)	3
Fairprene PS57-168 (Viton A, Dacron)	2
Fairprene (Viton A)	
5806	2
5807	2
5809	2
Kel-F-Dacron Diaphragm-VL-1101m4	4
Kel-F 5160 Diaphragm	2
Kel-F 5500 (gray) Diaphragm	2
Kel-F 5500 (gray) on Dacron diaphragm	2
Korda Flex (Teflon-coated glass fabric)	2
Silastic DC-9711 on Dacron diaphragm	3
Vinyl coated Fiberglass (gray-green)	3
9711 Silicone seal washer DC A 4094	3
adhesive (Dow Corning Silicate base)	
on aluminum	
9711 Silicone seal washer DC Chemloc	3
607 adhesive on aluminum	
<u>Porous Materials</u>	
Al-Si-Mag, Porous Ceramic No. 393	3
Aluminum Oxide, Porous-RA-98	2
Armalon-Teflon Felt (impregnated)	3
Armalon-	3
Dacron Cloth	2
Dac-2100	
Dac-2101	2
Dac-2102	2
Filtros C Stone (55 Micron)	2
Glass Cloth G-206-C	2
Poroloy--302SS Wire	3
Porous Kel-F (15-Micron Pore)	2
Porous Porcelain (1.4 Micron)	2
Porous Teflon (9-Micron Pore)	2
Rigimesh J SS, Wire	4
Sintered 300 Series SS Powder Compact	4
Sintered 302 SS Powder Compact	4
Sintered 316 SS Powder Compact	4
Sintered 302B SS Powder Compact	4
Sintered 316 and Cb SS Powder Compact	4
Teflon Cloth (25 Grade)	3
Teflon Cloth-Repeat (25 Grade)	2
Teflon Cloth (40 Grade)	3
Teflon Felt (impregnated)	
Teflon Cloth T-2300	3
Teflon Cloth T-2305	2

TABLE 28. (Continued)

Material	Classification	Impact Sensitive
<u>Lubricants</u>		
Alkaterge C	4	Yes
Amino Silane Oil and Grease	4	Yes
Apiezon Hardwax W	4	
Arochlor 1221	4	Yes
Arochlor 1232	4	Yes
Arochlor 1242	4	Yes
Arochlor 1248	4	Yes
Arochlor 1254	4	Yes
Bardahl	4	Yes
Carum 200	4	Yes
Ceresin Max	4	Yes
CFB-1	4	Yes
Dichloro-bis-tri-fluoromethyl benzene	3	No(c)
Dichlorohexafluorobutene	3	No(c)
Fluorolube FS	2	No(c)
Fluorolube FS plus 5% fluorolube light grease	2	No(c)
Fluorolube heavy grease 10214	2	No(c)
Fluorolube oil 10213	2	No(c)
Fluorolube S	2	No(c)
Fluorolube T	2	No(c)
Fluorolube Oil, S-30	2	No(d)
Fluorolube Grease, Hg-1200	2	No(d)
Fluorolube Grease, GR-660	2	No(d)
Formulation		Yes
F-9	4	
OS-16	4	Yes
OS-22	4	Yes
OS-23	4	Yes
OS-27	4	Yes
OS-28	4	Yes
OS-30	4	Yes
OS-32	4	Yes
OS-33	4	Yes
OS-34	4	Yes
OS-35	4	Yes
OS-37	4	Yes
CP-3898-2	4	Yes
Skydrol (uncolored)	4	Yes
Halocarbon Oil 8-25 AV	2	No(c)
Halocarbon Grease, Series 25-10	2	No(d)
Halocarbon Hi-Temp Stopcock Grease	2	No(d)
Halocarbon Oil 10-21	2	No(c)
Halocarbon Oil 11-14	2	No(c)
Halocarbon Stopcock Grease	2	No(c)
Hexachlorobutadiene	3	No(d)
Hexachloropropylene	4	No(c)
Hydraulic fluid RPM	4	
Hydraulic Oil Houghton Safe 620	3	Yes
Kel-F Alkane	2	
Kel-F Oil Cut No. 1	2	No(d)
Kel-F Oil No. 10	2	No(c)
Kel-F No. 90 Grease	2	No(c)
Lindol HF (tricresyl phosphate)	4	Yes
Lindol HFX	4	Yes
Liqui-Moly Concentrate	4	Yes
Lubri-Seal	4	Yes
Mineral Oil	4	Yes
Paraffin	4	Yes
Perfluorolube Grease FCD-759	2	Yes
Perfluorolube Oil FC-331	2	No(c)
FC-332	2	No(c)
FC-333	2	No(c)
FC-334	2	No(c)
FC-335	2	No(c)
Petrolatum	4	Yes
Polychloropentane (stabilized)	4	Yes
Renex	4	Yes
Silicone XF 224	4	Yes
Silicone Oil DC-7	4	Yes
DC-44	4	Yes
DC-200	4	Yes
DC-550	4	Yes
DC-701	4	Yes
DC-702	4	Yes
DC-710	4	Yes
Silicone Oil GE 2V3733	4	Yes
GE 51346	4	Yes
Tectyl	4	Yes
1,1,2,2, tetrafluoroethyl dodecylether	4	Yes
Tributyl Phosphate	4	Yes
Ucon Hydrolube U-4	4	Yes



TABLE 28. (Continued)

Material	Classification
<b>Ceramics, Refractories, and Miscellaneous</b>	
Agate (natural)	3
Agate (polished)	3
Al-Si-Mag (porcelain)	2
Alundum LA 116	2
Boron Nitride	4
Carboloy 44-A	4
Carboloy 55-A	4
Carboloy 78	4
Carboloy 999	4
Ceramic AB-2	2
Ceramic Al-200	2
Charcoal	4
Crystalor. (Silicon Carbide)	4
Graphitar No. 30	4
Graphite P5A6 Silver impregnated	4
Graphite P-55 Copper impregnated	4
Graphite P-59L Copper impregnated	4
Graphite P-692	4
Karbate	4
Norbide	2
Synthetic Sapphire (polished)	1

**Protective Coatings**

(A. Recommended for Long-Time Contact and Splash Resistance)

Teflon	1 (e)
Kel-F	1 (e)
Kel-F on 1060 Aluminum	1
Kel-F on 5234 Aluminum	1
Kel-F on 5652 Aluminum	1
Glass-lining (Clear) Light-Gray	1
Glass-lining (Cobalt) Cobalt-Colored Glass	1

(B. Recommended for Splash Resistance Service Only)

Tygon Paint 7286 TP-81-Clear	3 (e)
Tygon Paint 71253 TP-107B	3 (e)
Corrosite No. 521	3 (e)
Corrosite No. 551	3 (e)
Corrosite No. 581	3 (e)
Plastic Metal No. 22	3 (e)
Saran Rubber Q-1875	3 (e)
Mv-Type No. 150	3 (e)
Amercoat No. 1262	3 (e)
Hellex	3 (e)
P-5, Copolymer	3 (e)
Neolac Gray No. 8588	3 (e)
Steelcote Stainless Steel	3 (e)

(C. Not Recommended for 90% H<sub>2</sub>O<sub>2</sub> Service)

Geon Latex 31X	4 (e)
Flexcoat No. 1 Black	4 (e)
Lithgow LC-600 (Gray)	4 (e)
Amercoat Red	4 (e)
Prufcoat Medium Gray	4 (e)
Lithgow LC (600) (Brown)	4 (e)
Veloform F-10 CPP304	4 (e)
Cordo -255A	4 (e)
Cordo Plastic Coating (E-1 Resin + H-26 Activator)	4 (e)
Chromalloy	4 (e)
Unichrome Drum Lining B-124-17	4 (e)
Uclon, System E Coating	4 (e)
EX63B Paint	4 (e)
Sealer EC8C1 With Accelerator	4 (e)

TABLE 28. (Continued)

Material	Formulation	Classification	Impact Sensitive (f)
<b>Joint Sealing Compounds</b>			
"Althion" E	Polyethylene resin	1 (e)	Non-impact sensitive
Dispersite 1820	Butyl rubber in alkaline dispersion	2 (e)	Non-impact sensitive
Duxseal	Teflon	4 (e)	Non-sensitive Impact
Calmar CB Pipe Seal		4 (e)	Sensitive
Grane Thread Lub.	Oxidizable Oil	4 (e)	1 doubtful impact from 10 trials
Fel-Pro, C-5	Colloidal Copper	4 (e)	Impact sensitive 1 of 4 trials
Graphite Paste	Graphite dispersion	4 (e)	Non-impact sensitive
Goop (Blue)		4 (e)	Positive
Goop (Silver)		4 (e)	Positive
Cyl-Seal		4 (e)	Non-impact sensitive
Molybdenite Pipe Dope		4 (e)	Not tested
Permatex, Aviation Form A Gasket No. 3		4	Not tested
OS-18 Lubricant	Skydrol + 3% benton 34	4 (e)	Impact
Pecora Cpd.		4 (e)	Impact
Plastic Metal	No. 22-aluminum filler, latex base. No. 333-55 filler, latex base	4 (e)	Impact
Rutland Pipe Dope		4 (e)	Sensitive when met
Skydrol	Mixture Skydrol (P-801) and talc	4 (e)	Sensitive 2 of 3 trials
T-Film	Teflon-water dispersion	4 (e)	Impact sensitive
Weco No-Gel.	No. 50 metallic zinc base	4 (e)	Impact
X-Pardc		4 (e)	Non-impact sensitive
Kel-F Grease No. 90	Polychlorotrifluoroethylene		Non-sensitive
Tin Plating On Aluminum 6061		4 (e)	
Alcone Thread Lubricant		4 (e)	Not tested
Radco-Seal No. 15		4	Not tested

(a) Based on service experience.

(b) After 24-hour screening at 66 °C (150 °F).

(c) Non-impact sensitive to 1 kg at room temperature.

(d) Non-impact sensitive to 3 kg at room temperature.

(e) Tested at room temperature.

(f) One-kg impact at room temperature.

TABLE 29. COMPATIBILITY OF MATERIALS WITH METHYLENE CHLORIDE (CH<sub>2</sub>Cl<sub>2</sub>)

Material	Temperature, F								References
	Gas				Liquid				
	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4	
<b>Metals</b>									
1100 Aluminum			125				75	100	211
Copper			125		100		75		211
Br Bronze			125				75		211
Al Bronze			125				75		211
St Bronze			125		100		75		211
Red Brass			125				75		211
Yellow Brass			125				75		211
Mild Steel			125		75		75		211
Cast Iron							75		211
Mi-Resist							75		211
Si-Iron							75		211
410 Stainless Steel							75		211
430 Stainless Steel					100		75		211
304 Stainless Steel					75		75		211
316 Stainless Steel					75		75		211
Monel	175		212						211
Worthite	175		212						211
Durimet 20					100				211
Carpentony 20									211
A-Nickel			75			100	75		211
Monel			125			100	75		211
Inconel			75				75		211
Hastelloy B					100				127, 129, 211
Hastelloy C					100	100			127, 129, 211
Ni-O-nel					75				211
Lead							75		211
Gold							75		211
Platinum							75		211
Tantalum							75		211
Silver							75		211
<b>Nonmetals</b>									
Glass							75		211
Stoneware							75		211
Rubber								75	211
Asbestos					75				211
Graphite					75				211

TABLE 10. (Continued)

Material	Temperature, F.								Reference
	ASTM Type IIIA				ASTM Type IIIB				
	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4	
<b>Lubricants</b>									
Starfire Service Oil No. 1		75			212				149
Avalon 100					212				149
Avalon 100A					212				149
Avalon 1240					212				149
Dynalene Fluid					75				149
Crook No. 5	75				257				149
Crook No. 12					75				149
Carbellity								160	149
<b>Plastics</b>									
Pelton					75				223
Resin 30								75	223
Resin 40								75	223
M Styrene Sheet								75	223
Lexan								75	223
Kaplan 11								75	223
Bakelite								75	223
Nylon								75	223
Ritec								75	223
UPE 200C								75	223
UPE 214G								75	223
Viton A-MV								75	223
LD-234								75	223
Phenol-Formaldehyde				75					222
Phenol-Furfural				75					222
Furan Resins				75					222
Vinyl-Formaldehyde				75					222
Melamine-Formaldehyde				75					222
Aniline-Formaldehyde				75					222
Glyceryl-Phthalate				75					222
Epoxy Resins				75					222
Silicone Compounds				75					222
Calcium Compounds				75					222
Silicac				75					222
Inorganic Plastics				75					222
Vinyl Chloride-Acetate Copolymers				75					222
Vinyl Chloride Resins				75					222
Vinylidene Chloride Resins				75					222
Vinyl Formal Resins				75					222
Vinyl Butyral Resins				75					222
Vinyl Alcohol Resins				75					222
Polyvinyl Carbazate Resins				75					222
Alkyd Resins				75					222
Polyester Resins				75					222
Polyacrylic Ester Resins				75					222
Methyl Methacrylic Resins				75					222
Polystyrenes				75					222
Polyethylene				75					222
Polytetrafluoroethylene				75					222
Nylon				75					222
Silicone				75					222
Diallylphthalate				75					222
Modified Styrene				75					222
Acrylonitrile Rubber				75					222
Chlorinated Rubber				75					222
Hard Rubber				75					222
<b>Protective Coatings for Metals</b>									
Hycar				75					222
Hycar 22-2				75					222
Nukem No. 40				75					222
Neoprene				75					222
Priogard				75					222
Saranette				75					222
Bitumastic No. 50				75					222
Epon 7B-100				75					222
Novelox B-100-10				75					222
Epony				75					222
Modified Epoxy				75					222
Vinylidene Chloride				75					222
Vinyl Chloride				75					222
Exon 400 XRG	75								222
Porelon	75								222
Fluoriline 100	75								222
XB-270	75	75							222
Kel-F	75								222
Teflon	75								222
Vinyl Mastic	75								222
DEL Series A				75					222
Protect A	75								222
Lanette Fluoro B	75								222
Nerve-Kote Fluorinox	75								222
<b>Coatings for Nonmetals</b>									
Silicate Cement				75					222
Kel-F				75					222
Ketol 102-C	75								222
Phenoline 310		75							222
Concrete				75					222
Lanette Fluoro B	75								222
Nerve-Kote Fluorinox	75								222
Vinyl Mastic	75								222
Exon 400 XRG	75								222



TABLE 83. COMPATIBILITY OF MATERIALS WITH NITROGEN TETROXIDE (N<sub>2</sub>O<sub>4</sub>) (<0.2% MOISTURE)

Material	Temperature, F								References
	Gas				Liquid				
	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4	
410 Stainless Steel	65				100				169, 174
410 Stainless Steel (Welded)					65				169
416 Stainless Steel	65				65				169, 174
440C Stainless Steel					100				169, 245
302 Stainless Steel					100				169
303 Stainless Steel					100				169, 174
304 Stainless Steel	65			K2155	140				68, 169, 174, 220
304L Stainless Steel					145				17, 56, 169
304L Stainless Steel (Welded)					65				169
316 Stainless Steel	65				65				169
321 Stainless Steel	65				65				50, 169
321 Stainless Steel (Welded)					65				169
347 Stainless Steel	65				130				50, 169, 174
347 Stainless Steel (Welded)					65				169
A-286 (Annealed)					100				50, 169
A-286 (Aged)					60				50
AM-35C (Annealed)					60				169
AM-35C (Condition A)					65				169
17-4PH (W100C)					100				169
17-4PH (W100C)					100				169
17-7PH (TW50C)	60				100				169, 174
17-7PH (RW50C)					100				169
PH15-7 Mo (Condition A)					180				12
16-25-6 Lead					85				174
MoSibidem				K2112		80	80	80	174, 220
Magnesium, 100A	41	45			60	44			169, 174
Magnesium, AZ101C	61	65			60	65	50		169, 174
Magnesium, HM21A-T8					60				50, 174
Al-Nickel	60				65			75	169, 174
Nickel Electroplate					60				165
Electroless Nickel Plate	65				100				169, 174
A-Nickel (Welded)					65				169
Inconel	65				65			75	169, 174
Monel	65				65			75	169, 174
N-Monel					65			75	174
Ni-Span-C					60				169
Inconel X					75				174
Hastelloy Alloys					75				174
Niivar					60				50, 169
Platinum					75	75			21
Silver								75	174, 221
Silver Paste								50	169
Silver Solder								50	50, 169
Easy Flo 45								80	169, 245
St. Feel 55Co, 15Ag, 5Pb								80	245
Metco Hard Facing Alloy W							100	100	169
Metco Hard Facing Alloy 100							100	100	169
Metco Hard Facing Alloy 510							100	100	169
TE Spec Bzaze							80	245	
Microseal 100-1 on AMCO A Mg				100			100	100	169
Microseal 100-1 OD on AZ31C Mg				75			75	75	169
Microseal 100-1 on 2014-T6 Al	100				100		100	100	169
718 Bzaze 60Ni-20Al					65				169
Pure Tin Solder on 303 SS					65				169
Easy Flo Bzaze on 347 SS					65				169
AMS 4775 Microbrazz on 347 SS					65				169
Cu-60 Bzaze (Mo-Ni-Co) on 347 SS						100			169
Tantalum	65			K2021	75				68, 169, 174, 211, 220
Tin					80				174, 220
Tin Plate					60				169
Titanium, 6SA					100				169
Titanium, 7SA					105				17, 169, 220
Titanium, Alloy 67	65(6)				65				169, 174
Titanium, B122(2SA)					60				50, 169
Titanium, C122(2AV)					185				69, 174
Titanium, RC133AM	50(6)				150				169
Titanium, 6Al-4V					150				17, 50
Tungsten Carbide	65					65			169
Zinc								80	174, 220
Zinc Plate								50	169
Zirconium								75	220
FLUORIDES									
Teflon					75				50
Teflon FFE					75	60			169
Teflon FEP					100	60		60	23, 169
Teflon Graphite					75	50			23, 50, 169
Teflon PPS					75	50			50, 169
Teflon Asbestos					75				50, 169
Teflon Filled With Glass					80				23, 169
Teflon Filled With CaF <sub>2</sub>					80				23, 169
Teflon (Silica Primed)								75	23
FPE-Resit					75			60	50, 169
Armoform 7700					75	60			50, 169
Armoform 7700B					75	50			50, 169
Fluorobaktes					60				50, 169
Fluorogreen					60				50, 169

Material	Temperature, °F								References
	Gas				Liquid				
	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4	
Kel-F								75	50
Kel-F 300 (unplasticized)								60	169
Kel-F 300 (Annealed)								60	169
Genetron GC						80		160	23, 169
Genetron CCK-38					65			160	169
Genetron M-25					65			160	169
Teflone A						80		75	23, 169
Alcor 191					67				169
Polyethylene								75	23, 50
Polyethylene (low density)						60		60	169
Polyethylene (high density)						60		60	169
Polyethylene (irradiated)							60	60	50, 169
Raythane N (irradiated)					65			65	169
Polypropylene						60		60	50, 169
Nylon								65	50
Zytel 101								65	50
Copren 591								60	23, 169
Nylar								60	169
Saran								60	23, 50, 169
Synar					80			80	50, 169
Delrin								60	169
Luxon								60	50, 169
Tedlar					67			80	50, 169
Marlex 50								75	50, 169
MI Fox								75	23
Spondite								75	23
Kodapak II								75	23, 169
Mycon 20								60	169
H-File								60	169
Epon 1031 (with PMCA)								80	169
Plexiglas II								75	50
Plexiglas CR-39								75	50, 169
Plexiglas 55								75	50
Silicone Laminate								50	50, 169
Therocil Laminate						60		75	50, 169
Epon Laminate								50	50, 169
Polyester Laminate								60	50, 169
TM-1571 Laminate								75	50
Polyvinylchloride								75	50
Ultron						80		80	50, 169
Opalon 1219							60	60	169
Opalon 1220								60	169
Opalon 1221								60	169
Opalon 1222								60	169
W-plate								60	169
<b>Elastomers</b>									
Peristaltic 14 Extruded Polypropylene Rubber								65	169
Formula 132 (Ethylene Propylene Rubber, ILC)								67	169
E-42-2								65	169
E-422-1								65	169
X-7000-1 to 7 and 9 to 11								65	169
Parker 805-70								80	169
Parker 846-80								80	169
Parker 846-7								80	169
Parker KB-1235-10								80	169
E-14, 258								65	50, 169
Endic 551								65	169
Hycar 2202								65	50, 169
Stillman SR613-75								65	169
Stillman 11502-34								65	169
Stillman TC-4, 9, 1A								65	169
Precision 1390K20								70	169
Precision Formula 120								67	169
Precision Formula 121								67	169
Parker KV-1235-2								80	169
Parker XV-1235-4								80	169
Parker KB-1235-10								80	169
Trifluorohydrofluoroethylene tetrafluoroethylene (TFM-TFE)								80	169
Viton A								67	50, 169
Viton B								60	50, 169
Stillman Extruded Omni X FBE-4								67	169
Fackel V424-1								67	169
Parker 77-045								60	169
EX921-A70								60	169
Kel-F Elastomer								80	169
Kel-F-370C								75	50, 169
Kel-F-9500								60	169
Stillman Thicon Fluorel								65	169
LS 52								60	50, 169
LS 63								80	169
<b>Naturel Rubber</b>									
Buna N								60	50, 169
B-318-7								80	169
Chlorbutyl 500 (Silicone)								60	50, 169
Gerlock 22								65	50, 169
Gerlock 90C								65	50, 169
Neoprene								60	50, 169
Hypalon 20								75	50
Polybutadiene								75	50
Ty-lac 1650								75	50
Ty-lac 1640C								75	50
Ty-lac 1640								75	50
Penton								75	50
Chloroprene								75	50

	Temperature, F								
	Gels				Liquids				
Material	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4	References
<b>Lubricants</b>									
DC 11		67				67			169
DC 55						80			169
DC PL VAC		67				67			169
DC 150									169
Rayco-DC Grease					65				169
Nel-P 90								60	169
Polyglycol Oil						60			169
FX 45								67	169
Molykote Z						67			169
Dilube 703					60				169
Electrofilix 86-C							60		169
Rayco-30 Grease							60		169
Halecarbon Grease						67		60	169
Nordseal 147S and 421						60			169
Microseal 10C-1									169
PD-88					67				169
LOK Safe								67	169
Flake Graphite					60				169
Johns Manville No. 40					60				169
Graphiter 2,14,29,30, and 86								60	169
CCP-72					67				169
Furber P3M and PMH					67				169
Valve Seal A						A*			169
Aplon L						60			169
Fluorolube MDS60							60		169
Fluorothers G					60				169
<b>Sealants and Sealing Compounds</b>									
Reddy Tube 100	67				160				169
Reddy Tube 200	A*				160				169
Waterglass-Graphite					67				169
Vulcan A			67				67		169
Dry-Loc Sealant									169
Taffin Tape (Unstretched)					60				169
TR 1422								60	50, 169
RTV 20								60	50, 169
Epoxy 928								60	50, 169
Tetraplex P-43								60	50, 169
Proseal 703								60	50, 169
Palmerline S-556								60	50, 169
Crystat M & CF					60			60	50, 169
DC Reticos								75	50
Epoxy V1								75	50
Epoxy V11								75	50
Epoxy 422								75	50
Epoxy 501								75	50
Epoxy 101								75	50
Epoxy E7109								75	50
EO 105								75	50
EO 106								75	50
Resinox SC1008								75	50
Resinox SC1013								75	50
Reinhold FI20-15								75	50
US Polymeric Alloy								75	50
<b>Adhesives</b>									
Armstrong A-6								60	50, 169
EO 547								60	50, 169
HT 426								60	50, 169
3M-AF-10								60	169
A-5					60				169
Epoxy 422								60	5

(e) Personal communication.

(b) Titanium ignites under impact, but ignition does not spread.

TABLE 39. COMPATIBILITY OF MATERIALS WITH NITROGEN TRICHLORIDE CONTAINING 3.2% WATER<sup>(1)</sup>

Material	Temperature, °F								Reference
	Gas				Liquid				
	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4	
<b>Metals</b>									
Aluminum, 3003-H34									
Aluminum, 5005, Welded						15	70	120	
Carbon Steel, ASTM-A285, Grade C						15	70	120	
304L Stainless Steel									
904L-700					15				
					100				
					150				
Titanium, TA									
Titanium, 6Al-4V									
					100				
					150				

[illegible]

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195	The Production of Powder-Metallurgy Tungsten Sheet and Plate, July 20, 1964
196	Report on the Fourth Maraging-Steel Project Review, August 19, 1964
197	Electrodeposited, Electroless, and Anodized Coatings on Beryllium, September 1, 1964
198	Surface Damage in Machined Beryllium, January 4, 1965
199	Machining of Titanium Alloys, February 2, 1965
200	Summary of the Ninth Meeting of the Refractory Composites Working Group, March 3, 1965

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